## Séction - A (Physics)

1. Two reswurs of resistance, $100 \Omega$ and $200 \Omega$ are connected in parallel in an electrical circuit. The ratio of the thermal energy developed in $100 \Omega$ to that in $200 \Omega$ in a given time is :
(1) $4: 1$
(2) $1: 2$
(3) $2: 1$
(4) $1: 4$
2. Two hollow conducting spheres of radii $R_{1}$ and $R_{2}$ $\left(R_{1} \gg R_{2}\right)$ have equal charges. The potential would be:
(1) dependent on the material property of the sphere
(2) more on bigger sphere
(3) more on smaller sphere
(4) equal on both the spheres
3. When two monochromatic lights of frequency, $\nu$ and $\frac{v}{2}$ are incident on a photoelectric metal, their stopping potential becomes $\frac{\mathrm{V}_{\mathrm{s}}}{2}$ and $\mathrm{V}_{\mathrm{s}}$ respectively. The threshold frequency for this metal is:
(1) $\frac{3}{2} v$
(2) $2 v$
(3) $3 v$
(4) $\frac{2}{3} v$
4. As the temperature increases, the electrical resistance:
(1) decreases for conductors but increases for semiconductors
(2) increases for both conductors and semiconductors
(3) decreases for both conductors and semiconductors
(4) increases for conductors but decreases for semiconductors
5. If the initial tension on a stretched string is doubled, then the ratio of the initial and final speeds of a transverse wave along the string is :
(1) $1: 2$
(2) $1: 1$
(3) $\sqrt{2}: 1$
(4) $1: \sqrt{2}$
6. Match List-I with List-II :

List-I
(Electromagnetic waves)
(a) AM radio waves
(b) Microwaves
(c) Infrared radiations
(d) X-rays
(i) $10^{-10} \mathrm{~m}$
(ii) $10^{2} \mathrm{~m}$
(iii) $10^{-2} \mathrm{~m}$
(iv) $10^{-4} \mathrm{~m}$

List-II
(Wavelength)

Choose the correct answer from the options given below:
(1) (a)-(ii), (b) -(iii), (c) - (iv), (d) - (i)
(2) (a)-(iv), (b)- (iii), (c) - (ii), (d) - (i)
(3) (a)-(iii), (b) - (ii), (c) - (i), (d) -(iv)
(4) (a)-(iii), (b) - (iv), (c) - (ii), (d)-(i)
7. The ratio of the radius of gyration of a thin uniform disc about an axis passing through its centre and normal to its plane to the radius of gyration of the disc about its diameter is :
(1) $1: \sqrt{2}$
(2) $2: 1$
(3) $\sqrt{2}: 1$
(4) $4: 1$
8. A biconvex lens has radii of curvature, 20 cm each. If the refractive index of the material of the lens is 1.5 , the power of the lens is :
(1) infinity
(2) +2 D
(3) +20 D
(4) +5 D
9. The graph which shows the variation of the de Broglie wavelength ( $\lambda$ ) of a particle and its associated momentum ( $p$ ) is :
(1)

(2)

(3)

(4)

10. When light propagates through a material medium of relative permittivity $\epsilon_{\mathrm{r}}$ and relative permeability $\mu_{\mathrm{r}^{\prime}}$, he velocity of light, $v$ is given by : (c velocity of light in vacuum)
(1) $v=\frac{c}{\sqrt{\epsilon_{\mathrm{r}} \mu_{\mathrm{r}}}}$
(2) $\quad v=c$
(3) $v=\sqrt{\frac{\mu_{\mathrm{r}}}{\epsilon_{\mathrm{T}}}}$
(4) $v=\sqrt{\frac{\epsilon_{r}}{\mu_{r}}}$
11. In the given nuclear reaction, the element $X$ is :

$$
{ }_{11}^{22} \mathrm{Na} \rightarrow \mathrm{X}+\mathrm{e}^{+}+v
$$

(1) ${ }_{12}^{22} \mathrm{Mg}$
(2) ${ }_{11}^{23} \mathrm{Na}$
(3) ${ }_{10}^{23} \mathrm{Ne}$
(4) ${ }_{10}^{22} \mathrm{Ne}$
12. An ideal gas undergoes four different processes from the same initial state as shown in the figure below. Those processes are adiabatic, isothermal, isobaric and isochoric. The curve which represents the adiabatic process among $1,2,3$ and 4 is :

(1) 4
(2) 1
(3) 2
(4) 3
13. The ratio of the distances travelled by a freely falling body in the $1^{\text {st }}, 2^{\text {nd }}, 3^{\text {rd }}$ and $4^{\text {th }}$ second :
(1) $1: 1: 1: 1$
(2) $1: 2: 3: 4$
(3) $1: 4: 9: 16$
(4) $1: 3: 5: 7$
14. Two objects of mass 10 kg and 20 kg respectin R6 connected to the two ends of a rigid rod restively are 10 m with negligible mass. The distance of the center of mass of the system from the 10 kg mass is :
(1) 5 m
(2) $\frac{10}{3} \mathrm{~m}$
(3) $\frac{20}{3} \mathrm{~m}$
(4) 10 m
15. The energy that will be ideally radiated by a 100 kW transmitter in 1 hour is :
(1) $1 \times 10^{5} \mathrm{~J}$
(2) $36 \times 10^{7} \mathrm{~J}$
(3) $36 \times 10^{4} \mathrm{~J}$
(4) $36 \times 10^{5} \mathrm{~J}$
16. An electric lift with a maximum load of 2000 kg (lift + passengers) is moving up with a constant speed of $1.5 \mathrm{~ms}^{-1}$. The frictional force opposing the motion is 3000 N . The minimum power delivered by the motor to the lift in watts is: $\left(\mathrm{g}=10 \mathrm{~ms}^{-2}\right)$
(1) 23500
(2) 23000
(3) 20000
(4) 34500
17. A copper wire of length 10 m and radius $\left(10^{-2} / \sqrt{\pi}\right) \mathrm{m}$ has electrical resistance of $10 \Omega$. The current density in the wire for an electric field strength of $10(\mathrm{~V} / \mathrm{m})$ is :
(1) $10^{5} \mathrm{~A} / \mathrm{m}^{2}$
(2) $10^{4} \mathrm{~A} / \mathrm{m}^{2}$
(3) $10^{6} \mathrm{~A} / \mathrm{m}^{2}$
(4) $10^{-5} \mathrm{~A} / \mathrm{m}^{2}$
18. If a soap bubble expands, the pressure inside the bubble:
(1) is equal to the atmospheric pressure
(2) decreases
(3) increases
(4) remains the same
19. Given below are two statements :

Statement I:
Biot-Savart's law gives us the 'expression for the magnetic field strength of an infinitesimal current element (Idl) of a current carryinig conductor only.
Statement II:
Biot-Savart's law is analogous to Coulomb's inverse square law of charge $q$, with the former being related to the field produced by a scalar source, Idl while the latter being produced by a vector source, $q$.
In light of above statements choose the most appropriate answer from the options given below:
(1) Statement I is incorrect and Statement II is correct
(2) Both Statement I and Statement II are correct
(3) Both Statement I and Statement II are incorrect
(4) Statement I is correct and Statement II is incorrect
20. A shell of mass $m$ is at rest initially. It explodes into three fragments having mass in the ratio $2: 2: 1$. If the fragments having equal mass fly off along mutually perpendicular directions with speed $v$, the speed of the third (lighter) fragment is :
(1) $3 \sqrt{2} v$
(2) $v$
(3) $\sqrt{2} v$
(4) $2 \sqrt{2} v$
21. A spherical ball is dropped in a long column of a highly viscous liquid. The curve in the graph shown, which represents the speed of the ball $(v)$ as a function of time $(t)$ is :

(1) D
(2) A
(3) B
(4) C
22. The angular speed of a fly wheel moving with uniform angular acceleration changes from 1200 rpm to 3120 rpm in 16 seconds. The angular acceleration in $\mathrm{rad} / \mathrm{s}^{2}$ is :
(1) $104 \pi$
(2) $2 \pi$
(3) $4 \pi$
(4) $12 \pi$
23. The peak voltage of the ac source is equal to :
(1) $1 / \sqrt{2}$ times the rms value of the ac source
(2) the value of voltage supplied to the circuit
(3) the rms value of the ac source
(4) $\sqrt{2}$ times the rms value of the ac source
24. The dimensions [ $\mathrm{MLT}^{-2} \mathrm{~A}^{-2}$ ] belong to the:
(1) electric permittivity
(2) magnetic flux
(3) self inductance
(4) magnetic permeability
25. The displacement-time graphs of two moving particles make angles of $30^{\circ}$ and $45^{\circ}$ with the $x$-axis as shown in the figure. The ratio of their respective velocity is :

(1) $1: \sqrt{3}$
(2) $\sqrt{3}: 1$
(3) $1: 1$
(4) $1: 2$
26.

(a)

(b)


In the given circuits (a), (b) and (c), the potential drop across the two $p-n$ junctions are equal in :
(1) Both circuits (a) and (c)
(2)
(3) Circuit (b) only
(4) Circuit (c) only
27. The angle between the electric lines of force and the equipotential surface is :
(1) $180^{\circ}$
(2) $0^{\circ}$
(3) $45^{\circ}$
(4) $90^{\circ}$
28. Plane angle and solid angle have :
(1) Both units and dimensions
(2) Units but no dimensions
(3) Dimensions but no units
(4) No units and no dimensions
29. In a Young's double slit experiment, a student observes 8 fringes in a certain segment of screen when a monochromatic light of 600 nm wavelength is used. If the wavelength of light is changed to 400 nm , then the number of fringes he would observe in the same region of the screen is:
(1) 12
(2) 6
(3) 8
(4) 9
30. A light ray falls on a glass surface of refractive index $\sqrt{3}$, at an angle $60^{\circ}$. The angle between the refracted and reflected rays would be :
(1) $120^{\circ}$
(2) $30^{\circ}$
(3) $60^{\circ}$
(4) $90^{\circ}$
31. In half wave rectification, if the input frequency is 60 Hz , then the output frequency would be:
(1) 120 Hz
(2) zero
(3) 30 Hz
(4) 60 Hz
32. A body of mass 60 g experiences a gravitational force of 3.0 N , when placed at a particular point. The magnitude of the gravitational field intensity at that point is:
(1) $180 \mathrm{~N} / \mathrm{kg}$
(2) $0.05 \mathrm{~N} / \mathrm{kg}$
(3) $50 \mathrm{~N} / \mathrm{kg}$
(4) $20 \mathrm{~N} / \mathrm{kg}$
33. A square loop of side 1 m and resistance $1 \Omega$ is placed in a magnetic field of 0.5 T . If the plane of loop is perpendicular to the direction of magnetic field, the magnetic flux through the loop is :
(1) zero weber
(2) 2 weber
(3) 0.5 weber
(4) 1 weber
34. Let $T_{1}$ and $T_{2}$ be the energy of an electron in the first and second excited states of hydrogen atom, respectively. According to the Bohr's model of an atom, the ratio $T_{1}: T_{2}$ is :
(1) $9: 4$
(2) $1: 4$
(3) $4: 1$
(4) $4: 9$
35. A long solenoid of radius 1 mm has 100 turns per mm . If 1 A current flows in the solenoid, the magnetic field strength at the centre of the solenoid is :
(1) $6.28 \times 10^{-4} \mathrm{~T}$
(2) $6.28 \times 10^{-2} \mathrm{~T}$
(3) $12.56 \times 10^{-2} \mathrm{~T}$
(4) $12.56 \times 10^{-4} \mathrm{~T}$

## Section - B (Physics)

36. Two pendulums of length 121 cm and 100 cm start vibrating in phase. At some instant, the two are at their mean position in the same phase. The minimum number of vibrations of the shorter pendulum after which the two are again in phase at the mean position is :
(1) 8
(2) 11
(3) 9
(4) 10
37. The volume occupied by the molecules contained in 4.5 kg water at STP, if the intermolecular forces vanish away is :
(1) $5.6 \mathrm{~m}^{3}$
(2) $5.6 \times 10^{6} \mathrm{~m}^{3}$
(3) $5.6 \times 10^{3} \mathrm{~m}^{3}$
(4) $5.6 \times 10^{-3} \mathrm{~m}^{3}$
38. Two transparent media $A$ and $B$ are separated by a plane boundary. The speed of light in those media are $1.5 \times 10^{8} \mathrm{~m} / \mathrm{s}$ and $2.0 \times 10^{8} \mathrm{~m} / \mathrm{s}$, respectively. The critical angle for a ray of light for these two media
is :
(1) $\tan ^{-1}(0.750)$
(2) $\sin ^{-1}(0.500)$
(3) $\sin ^{-1}(0.750)$
(4) $\tan ^{-1}(0.500)$
39. Match List - I with List - II :

## List - I

(a) Gravitational constant (G)
(b) Gravitational potential energy
(c) Gravitational potential
(d) Gravitational intensity
(i) $\begin{aligned} & \text { List - II } \\ & {\left[\mathrm{L}^{2} \mathrm{~T}^{-2}\right]}\end{aligned}$
(ii) $\left[\mathrm{M}^{-1} \mathrm{~L}^{3} \mathrm{~T}^{-2}\right]$
(iii) $\left[\mathrm{LT}^{-2}\right]$
(iv) $\left[\mathrm{ML}^{2} \mathrm{~T}^{-2}\right]$

Choose the correct answer from the options given below:
(1)
(2) (a) - (iv), (b) - (ii), (c) - (i), (d) - (iii)
(3) (a) - (ii), (b) - (i), (c) - (iv), (d) - (iii)
(4) (a) - (ii), (b) - (iv), (c) - (i), (d) - (iii)
(4) (a) - (ii), (b) - (iv), (c) - (iii), (d) - (i)
40. A series LCR circuit with inductance 10 H , capacitance $10 \mu \mathrm{~F}$, resistance $50 \Omega$ is connected to an ac source of voltage, $V=200 \sin (100 t)$ volt. If the resonant frequency of the LCR circuit is $v_{\mathrm{o}}$ and the frequency of the ac source is $v$, then :
(1) $v=100 \mathrm{~Hz} ; v_{0}=\frac{100}{\pi} \mathrm{~Hz}$
(2) $v_{\mathrm{o}}=v=50 \mathrm{~Hz}$
(3) $v_{0}=v=\frac{50}{\pi} \mathrm{~Hz}$
(4)

$$
v_{o}=\frac{50}{\pi} \mathrm{~Hz}, \nu=50 \mathrm{~Hz}
$$

41. A ball is projected with a velocity, $10 \mathrm{~ms}^{-1}$, at an angle of $60^{\circ}$ with the vertical direction. Its speed at the highest point of its trajectory will be:
(1) $10 \mathrm{~ms}^{-1}$
(2) Zero
(3) $5 \sqrt{3} \mathrm{~ms}^{-1}$
(4) $5 \mathrm{~ms}^{-1}$
42. From Ampere's circuital law for a long straight wire of circular cross-section carrying a steady current, the variation of magnetic field in the inside and outside region of the wire is :
(1) a linearly decreasing function of distance upto the boundary of the wire and then a linearly increasing one for the outside region.
(2) uniform and remains constant for both the regions.
(3) a linearly increasing function of distance upto the boundary of the wire and then linearly decreasing for the outside region.
(4) a linearly increasing function of distance $r$ upto the boundary of the wire and then decreasing one with $1 / \mathrm{r}$ dependence for the outside region.
43. Given below are two statements: One is labelled as Assertion (A) and the other is labelled as Reason (R).

## Assertion (A) :

The stretching of a spring is determined by the shear modulus of the material of the spring.

## Reason (R) :

A coil spring of copper has more tensile strength than a steel spring of same dimensions.
In the light of the above statements, choose the most appropriate answer from the options given below :
(1) (A) is false but ( $R$ ) is true
(2) Both (A) and (R) are true and (R) is the correct explanation of (A)
(3) Both (A) and (R) are true and (R) is not the correct explanation of (A)
(4) (A) is true but (R) is false
44.


The truth table for the given logic circuit is :

| A | B | C |
| :---: | :---: | :---: |
| 0 | 0 | 0 |
| 0 | 1 | 1 |
| 1 | 0 | 0 |
| 1 | 1 | 1 |

(2)

| A | B | C |
| :---: | :---: | :---: |
| 0 | 0 | 0 |
| 0 | 1 | 1 |
| 1 | 0 | 1 |
| 1 | 1 | 0 |

(3)

| A | B | C |
| :---: | :---: | :---: |
| 0 | 0 | 1 |
| 0 | 1 | 0 |
| 1 | 0 | 0 |
| 1 | 1 | 1 |


| A | B | C |
| :---: | :---: | :---: |
| 0 | 0 | 1 |
| 0 | 1 | 0 |
| 1 | 0 | 1 |
| 1 | 1 | 0 |

45. A nucleus of mass number 189 splits into two nuclei having mass number 125 and 64 . The ratio of radius of two daughter nuclei respectively is :
(1) $25: 16$
(2) $1: 1$
(3) $4: 5$
(4) $5: 4$
46. A wheatstone bridge is used to determine the value of unknown resistance $X$ by adjusting the variable resistance $Y$ as shown in the figure. For the most precise measurement of $X$, the resistances $P$ and $Q$ :

(1) do not play any significant role
(2) should be approximately equal to $2 X$
(3) should be approximately equal and are small (4) should be very large and unequal
47. The area of a rectangular field (in $\mathrm{m}^{2}$ ) of length 55.3 m and breadth 25 m after rounding off the value for correct significant digits is :
(1) $14 \times 10^{2}$
(2) $138 \times 10^{1}$
(3) 1382
(4) 1382.5
48. A big circular coil of 1000 turns and average radius 10 m is rotating about its horizontal diameter at $2 \mathrm{rad} \mathrm{s}{ }^{-1}$. If the vertical component of earth's magnetic field at that place is $2 \times 10^{-5} \mathrm{~T}$ and electrical resistance of the coil is $12.56 \Omega$, then the maximum induced current in the coil will be :
```
(1) 2 A
(2) }0.25\textrm{A
(3) }1.5\textrm{A
(4) }1\textrm{A
```

49. Two point charges -q and +q are placed at a distance of L , as shown in the figure.


The magnitude of electric field intensity at a distance $R(R \gg L)$ varies as :
(1) $\frac{1}{\mathrm{R}^{6}}$
(2) $\frac{1}{R^{2}}$
(3) $\frac{1}{\mathrm{R}^{3}}$
(4) $\frac{1}{\mathrm{R}^{4}}$
50. A capacitor of capacitance $\mathrm{C}=900 \mathrm{pF}$ is charged fully by 100 V battery B as shown in figure (a). Then it is disconnected from the battery and connected to another uncharged capacitor of capacitance $\mathrm{C}=900 \mathrm{pF}$ as shown in figure (b). The electrostatic energy stored by the system (b) is :
(a)

(b)

(1) $1.5 \times 10^{-6} \mathrm{~J}$
(2) $4.5 \times 10^{-6} \mathrm{~J}$
(3) $3.25 \times 10^{-6} \mathrm{~J}$
(4) $2.25 \times 10^{-6} \mathrm{~J}$

## Section-A (Chemistry)

51. Which statement regarding polymers is not correct?
(1) Thermosetting polymers are reusable.
(2) Elastomers have polymer chains held together by weak intermolecular forces.
(3) Fibers possess high tensile strength.
(4) Thermoplastic polymers are capable of repeatedly softening and hardening on heating and cooling respectively.
52. At 298 K , the standard electrode potentials of $\mathrm{Cu}^{2+}$ / $\mathrm{Cu}, \mathrm{Zn}^{2+} / \mathrm{Zn}, \mathrm{Fe}^{2+} / \mathrm{Fe}$ and $\mathrm{Ag}^{+} / \mathrm{Ag}$ are 0.34 V , $-0.76 \mathrm{~V},-0.44 \mathrm{~V}$ and 0.80 V , respectively.
On the basis of standard electrode potential, predict
which of the following reaction can not occur?
(1) $\quad 2 \mathrm{CuSO}_{4}(\mathrm{aq})+2 \mathrm{Ag}(\mathrm{s}) \rightarrow 2 \mathrm{Cu}(\mathrm{s})+\mathrm{Ag}_{2} \mathrm{SO}_{4}(\mathrm{aq})$
(2) $\mathrm{CuSO}_{4}(\mathrm{aq})+\mathrm{Zn}(\mathrm{s}) \rightarrow \mathrm{ZnSO}_{4}(\mathrm{aq})+\mathrm{Cu}(\mathrm{s})$
(3) $\mathrm{CuSO}_{4}(\mathrm{aq})+\mathrm{Fe}(\mathrm{s}) \rightarrow \mathrm{FeSO}_{4}(\mathrm{aq})+\mathrm{Cu}(\mathrm{s})$
(4) $\mathrm{FeSO}_{4}(\mathrm{aq})+\mathrm{Zn}(\mathrm{s}) \rightarrow \mathrm{ZnSO}_{4}(\mathrm{aq})+\mathrm{Fe}(\mathrm{s})$
53. The IUPAC name of an element with atomic number 119 is
(1) ununoctium
(2) ununennium
(3) unnilennium
(4) unununnium
54. Given below are two statements :

Statement I:
In the coagulation of a negative sol, the flocculating power of the three given ions is in the order -

$$
\mathrm{Al}^{3+}>\mathrm{Ba}^{2+}>\mathrm{Na}^{+}
$$

Statement II:
In the coagulation of a positive sol, the flocculating power of the three given salts is in the order $\mathrm{NaCl}>\mathrm{Na}_{2} \mathrm{SO}_{4}>\mathrm{Na}_{3} \mathrm{PO}_{4}$
In the light of the above statements, choose the most appropriate answer from the options given below :
(1) Statement I is incorrect but Statement II is
(2) Both Statement I and Statement II are correct.
(3) Both Statement I and Statement II are
(4) Statement I is correct but Statement II is incorrect.
55. Which of the following statement is not correct about diborane?
(1) Both the Boron atoms are $s p^{2}$ hybridised.
(2) There are two 3-centre-2-electron bonds.
(3) The four terminal B-H bonds are two centre
(4) two electron bonds.
(4) The four terminal Hydrogen atoms and the two Boron atoms lie in one plane.
56. $\mathrm{RMgX}+\mathrm{CO}_{2} \xrightarrow[\text { ether }]{\text { dry }} \mathrm{Y} \xrightarrow{\mathrm{H}_{3} \mathrm{O}^{+}} \mathrm{RCOOH}$

What is Y in the above reaction?
(1) $(\mathrm{RCOO})_{2} \mathrm{Mg}$
(2) $\mathrm{RCOO}^{-} \mathrm{Mg}^{+} \mathrm{X}$
(3) $\mathrm{R}_{3} \mathrm{CO}^{-} \mathrm{Mg}^{+} \mathrm{X}$
(4) $\mathrm{RCOO}^{-} \mathrm{X}^{+}$
57. What mass of $95 \%$ pure $\mathrm{CaCO}_{3}$ will be required to neutralise 50 mL of 0.5 M HCl solution according to the following reaction?
$\left[\mathrm{CaCO}_{3(\mathrm{~s})}+2 \mathrm{HCl}_{(\mathrm{aq})} \rightarrow \mathrm{CaCl}_{2(\mathrm{aq})}+\mathrm{CO}_{2(\mathrm{~g})}+2 \mathrm{H}_{2} \mathrm{O}_{(\mathrm{l})}\right.$
[Calculate upto second place of decimal point]
(1) 9.50 g
(2) 1.25 g
(3) 1.32 g
(4) 3.65 g
58. Which amongst the following is incorrect statement?
(1) $\mathrm{O}_{2}^{+}$ion is diamagnetic.
(2) The bond orders of $\mathrm{O}_{2}^{+}, \mathrm{O}_{2}, \mathrm{O}_{2}^{-}$and $\mathrm{O}_{2}^{2-}$
(3) $\mathrm{C}_{2}$ molecule $2,1.5$ and 1 , respectively. degenerate $\pi$ molecular electrons in its two
(4) $\mathrm{H}_{2}^{+}$ion has one electron.
59. Amongst the following which one will have
maximum 'lone pair - lone pair' electron repulsions?
$\begin{array}{ll}\text { (1) } & \mathrm{XeF}_{2} \\ \text { (2) } & \mathrm{ClF}_{3}\end{array}$
(3) $\quad \mathrm{IF}_{5}$
(4) $\mathrm{SF}_{4}$
60. Choose the correct statement:
(1) Both diamond and graphite are used as dry
(2) Diamond and graphite have two dimensional
network.
(3) Diamond is covalent and graphite is ionic.
(4) Diamond is $s p^{3}$ hybridised and graphite is
61. Given below are two statements : one is labelled as Assertion (A) and the other is labelled as Reason
(R). Assertion (A) :
In a particular point defect, an ionic solid is electrically neutral, even if few of its cations are missing from its unit cells.

## Reason (R):

In an ionic solid, Frenkel defect arises due to dislocation of cation from its lattice site to interstitial site, maintaining overall electrical neutrality.
In the light of the above statements, choose the most
appropriate answer from the options given below:
(1) (A) is not correct but (R) is correct
(2) Both (A) and (R) are correct and (R) is the
(3) Borrect explanation of (A)
(3) Both (A) and (R) are correct but (R) is not the correct explanation of (A)
(4) (A) is correct but $(R)$ is not correct
62. Match List-I with List - II.

## List - I

| (a) | Li | (i) | List-II |
| :--- | :--- | :--- | :--- |
| absorbent for carbon dioxide |  |  |  |
| (b) | Na | (ii) | electrochemical cells |
| (c) | KOH | (iii) | coolant in fast breeder reactors |
| (d) | Cs | (iv) | photoelectric cell | Choose the correct answer from the options given

below:
(1)
(a) - (ii), (b) - (iii), (c) - (i), (d) - (iv)
(2) (a)-(iv), (b)-(i), (c) -(iii), (d)-(ii)
(3) (a) - (iii), (b) - (iv), (c) - (ii), (d) -(i)
(4) (a)-(i),(b)-(iii), (c) - (iv), (d)-(ii)
63. Given below are half cell reactions :
$\mathrm{MnO}_{4}^{-}+8 \mathrm{H}^{+}+5 \mathrm{e}^{-} \rightarrow \mathrm{Mn}^{2+}+4 \mathrm{H}_{2} \mathrm{O}$,
$\mathrm{E}_{\mathrm{Mn}^{2+} / \mathrm{MnO}_{4}^{-}}^{\circ}=-1.510 \mathrm{~V}$
$\frac{1}{2} \mathrm{O}_{2}+2 \mathrm{H}^{+}+2 \mathrm{e}^{-} \rightarrow \mathrm{H}_{2} \mathrm{O}$,
$\mathrm{E}_{\mathrm{O}_{2} / \mathrm{H}_{2} \mathrm{O}}^{\circ}=+1.223 \mathrm{~V}$
Will the permanganate ion, $\mathrm{MnO}_{4}^{-}$liberate $\mathrm{O}_{2}$ from water in the presence of an acid ?
(1) No, because $\mathrm{E}_{\text {cell }}^{\circ}=-2.733 \mathrm{~V}$
(2) Yes, because $\mathrm{E}_{\text {cell }}^{\circ}=+0.287 \mathrm{~V}$
(3) No, because $E_{\text {cell }}^{\circ}=-0.287 \mathrm{~V}$
(4) Yes, because $\mathrm{E}_{\text {cell }}^{\circ}=+2.733 \mathrm{~V}$
64. Match List - I with List - II.

List-I
(Hydrides)
(a) $\mathrm{MgH}_{2}$
(b) $\mathrm{GeH}_{4}$
(c) $\mathrm{B}_{2} \mathrm{H}_{6}$
(d) HF

## List - II

(Nature)
(i) Electron precise
(ii) Electron deficient
(iii) Electron rich
(iv) Ionic

Choose the correct answer from the options given below:
(a) - (ii), (b) - (iii), (c) - (iv), (d) - (i)
(a) - (iv), (b) - (i), (c) - (ii), (d) - (iii)
(3) (a) - (iii), (b) - (i), (c) - (ii), (d) - (iv)
(4)
(a) - (i), (b) - (ii), (c) - (iv), (d) - (iii)
65. Given below are two statements :

## Statement I:

The acidic strength of monosubstituted nitrophenol is higher than phenol because of electron withdrawing nitro group.

## Statement II :

0 -nitrophenol, $m$-nitrophenol and $p$-nitrophenol will have same acidic strength as they have one nitro group attached to the phenolic ring.
In the light of the above statements, choose the most appropriate answer from the options given below :
(1) Statement I is incorrect but Statement II is correct.
(2) Both Statement I and Statement II are correct.
(3) Both Statement I and Statement II are incorrect.
(4) Statement I is correct but Statement II is incorrect.
66. Match List - I with List - II.

List - I
(Drug class)
(a) Antacids

List - II
(Drug molecule)
(b) Antihistamines
(c) Analgesics
(i) Salvarsan
(ii) Morphine
(iii) Cimetidine
(iv) Seldane
(d) Antimicrobials

Choose the correct answer from the options given
below:
(2) (a) - (iii), (b) - (ii), (c) - (iv), (d) - (i)
(3) (a) - (iii), (b) - (iv), (c) - (ii), (d) - (i)
(a) - (i), (b) - (iv), (c) - (ii), (d) - (iii)
67. The incorrect statement regarding enzymes is :
(1) Enzymes are very specific for a particular reaction and substrate.
(2) Enzymes are biocatalysts.
(3) Like chemical catalysts enzymes reduce the activation energy of bio processes.
(4) Enzymes are polysaccharides.
68. Identify the incorrect statement from the following.
(1) The shapes of $d_{x y}, d_{y z^{\prime}}$ and $d_{z x}$ orbitals are similar to each other; and $d_{x^{2}-y^{2}}$ and $d_{z} 2$ are similar to each other.
(2) All the five $5 d$ orbitals are different in size when compared to the respective $4 d$ orbitals.
(3) All the five $4 d$ orbitals have shapes similar to the respective $3 d$ orbitals.
(4) In an atom, all the five $3 d$ orbitals are equal in energy in free state.
69. The incorrect statement regarding chirality is :
(1) A racemic mixture shows zero optical rotation.
(2) $\mathrm{S}_{\mathrm{N}} 1$ reaction yields $1: 1$ mixture of both enantiomers.
(3) The product obtained by $S_{N} 2$ reaction of haloalkane having chirality at the reactive site shows inversion of configuration.
(4) Enantiomers are superimposable mirror images on each other.
63. Given below are half cell reactions:
$\mathrm{MnO}_{4}^{-}+8 \mathrm{H}^{+}+5 \mathrm{e}^{-} \rightarrow \mathrm{Mn}^{2+}+4 \mathrm{H}_{2} \mathrm{O}$,
$\mathrm{E}_{\mathrm{Mn}^{2+} / \mathrm{MnO}_{4}^{-}}^{\circ}=-1.510 \mathrm{~V}$
$\frac{1}{2} \mathrm{O}_{2}+2 \mathrm{H}^{+}+2 \mathrm{e}^{-} \rightarrow \mathrm{H}_{2} \mathrm{O}$,
$\mathrm{E}_{\mathrm{O}_{2} / \mathrm{H}_{2} \mathrm{O}}^{\circ}=+1.223 \mathrm{~V}$
Will the permanganate ion, $\mathrm{MnO}_{4}^{-}$liberate $\mathrm{O}_{2}$ from water in the presence of an acid ?
(1) No, because $\mathrm{E}_{\text {cell }}^{\circ}=-2.733 \mathrm{~V}$
(2) Yes, because $\mathrm{E}_{\text {cell }}^{\circ}=+0.287 \mathrm{~V}$
(3) No, because $\mathrm{E}_{\text {cell }}^{\circ}=-0.287 \mathrm{~V}$
(4) Yes, because $\mathrm{E}_{\text {cell }}^{\circ}=+2.733 \mathrm{~V}$
64. Match List - I with List - II.

## List - I

(Hydrides)
(a) $\mathrm{MgH}_{2}$
(b) $\mathrm{GeH}_{4}$
(c) $\mathrm{B}_{2} \mathrm{H}_{6}$
(d) HF

Choose the correct answer from the options given below :
(1) (a) - (ii), (b) - (iii), (c) - (iv), (d) - (i)
(2) (a) - (iv), (b) - (i), (c) - (ii), (d) - (iii)
(3) (a) - (iii), (b) - (i), (c) - (ii), (d) - (iv)
(4) (a) - (i), (b) - (ii), (c) - (iv), (d) - (iii)
65. Given below are two statements :

## Statement I:

The acidic strength of monosubstituted nitrophenol is higher than phenol because of electron withdrawing nitro group.

## Statement II :

$o$-nitrophenol, $m$-nitrophenol and $p$-nitrophenol will have same acidic strength as they have one nitro group attached to the phenolic ring.
In the light of the above statements, choose the most appropriate answer from the options given below :
(1) Statement I is incorrect but Statement II is correct.
(2) Both Statement I and Statement II are correct.
(3) Both Statement I and Statement II are incorrect.
(4) Statement I is correct but Statement II is incorrect.
(iv) Ionic

## List - II

(Nature)
(i) Electron precise
(ii) Electron deficient
(iii) Electron rich
66. Match List - I with List - II.

List - I
(Drug class)
List-II
(Drug molecule)
(a) Antacids
(b) Antihistamines
(c) Analgesics
(d) Antimicrobials
(i) Salvarsan
(ii) Morphine
(iii) Cimetidine
(iv) Seldane

Choose the correct answer from the options given below:
(1) (a) - (iv), (b) - (iii), (c) - (i), (d) - (ii)
(2) (a) - (iii), (b) - (ii), (c) - (iv), (d) - (i)
(3) (a) - (iii), (b) - (iv), (c) - (ii), (d) - (i)
(a) - (i), (b) - (iv), (c) - (ii), (d) - (iii)
67. The incorrect statement regarding enzymes is :
(1) Enzymes are very specific for a particular reaction and substrate.
(2). Enzymes are biocatalysts.
(3) Like chemical catalysts enzymes reduce the activation energy of bio processes.
(4) Enzymes are polysaccharides.
68. Identify the incorrect statement from the following.
(1) The shapes of $d_{x y}, d_{y z}$ and $d_{z x}$ orbitals are similar to each other ; and $d_{x}{ }^{2}-y^{2}$ and $d_{z}{ }^{2}$ are similar to each other.
(2) All the five $5 d$ orbitals are different in size
(3) All the five $4 d$ orbitals have shapes similar to the respective $3 d$ orbitals.
(4) In an atom, all the five $3 d$ orbitals are equal in energy in free state.
69. The incorrect statement regarding chirality is :
(1) A racemic mixture shows zero optical rotation.
(2) $\mathrm{S}_{\mathrm{N}} 1$ reaction yields 1:1 mixture of both enantiomers.
(3) The product obtained by $\mathrm{S}_{\mathrm{N}} 2$ reaction of haloalkane having chirality at the reactive site shows inversion of configuration.
(4) Enantiomers are superimposable mirror images on each other.
70. Which compound amongst the following is not an aromatic compound ?
(1)

(2)

(3)

(4)

71. The given graph is a representation of kinetics of a reaction.


The $y$ and $x$ axes for zero and first order reactions, respectively are
(1) zero order ( $y=$ rate and $x=$ concentration), first order ( $y=$ rate and $x=\mathrm{t}_{1 / 2}$ )
(2) zero order ( $y=$ concentration and $x=$ time), first order $\left(y=\mathrm{t}_{1 / 2}\right.$ and $x=$ concentration)
(3) zero order ( $y=$ concentration and $x=$ time), first order ( $y=$ rate constant and $x=$ concentration)
(4) zero order ( $y=$ rate and $x=$ concentration), first order $\left(y=\mathrm{t}_{1 / 2}\right.$ and $x=$ concenration $)$
72. Which one is not correct mathematical equation for Dalton's Law of partial pressure? Here $p=$ total pressure of gaseous mixture
(1) $\mathrm{p}_{i}=\chi_{i} \mathrm{p}_{i}^{\mathrm{o}}$, where $\chi_{i}=$ molefraction of $i^{\text {th }}$ gas in gaseous mixture $\mathrm{p}_{i}^{\mathrm{o}}=$ pressure of $i^{\text {th }} \mathrm{gas}$ in pure state
(2) $\mathrm{p}=\mathrm{p}_{1}+\mathrm{p}_{2}+\mathrm{p}_{3}$
(3) $\mathrm{p}=\mathrm{n}_{1} \frac{\mathrm{RT}}{\mathrm{V}}+\mathrm{n}_{2} \frac{\mathrm{RT}}{\mathrm{V}}+\mathrm{n}_{3} \frac{\mathrm{RT}}{\mathrm{V}}$
(4) $\mathrm{p}_{i}=\chi_{i} \mathrm{p}$, where $\mathrm{p}_{i}=$ partial pressure of $i^{\text {th }}$ gas
$\chi_{i}=$ mole fraction of $i^{\text {th }}$ gas in gaseous mixture
73. Given below are two statements :

## Statement I:

Primary aliphatic amines react with $\mathrm{HNO}_{2}$ to give unstable diazonium salts.

## Statement II:

Primary aromatic amines react with $\mathrm{HNO}_{2}$ to form diazonium salts which are stable even above 300 K .
In the light of the above statements, choose the most appropriate answer from the options given below:
(1) Statement I is incorrect but Statement II is correct.
(2) Both Statement I and Statement II are correct.
(3) Both Statement I and Statement II are incorrect.
(4) Statement I is correct but Statement II is incorrect.
74. Identify the incorrect statement from the following
(1) Lithium is the strongest reducing agentamong the alkali metals.
(2) Alkali metals react with water to form their hydroxides.
(3) The oxidation number of K in $\mathrm{KO}_{2}$ is +4 .
(4) Ionisation enthalpy of alkali metals decreases from top to bottom in the group.
75. The IUPAC name of the complex-
$\left[\mathrm{Ag}\left(\mathrm{H}_{2} \mathrm{O}\right)_{2}\right]\left[\mathrm{Ag}(\mathrm{CN})_{2}\right]$ is :
(1) diaquasilver(I) dicyanidoargentate(I)
(2) dicyanidosilver(II) diaquaargentate(II)
(3) diaquasilver(II) dicyanidoargentate(II)
(4) dicyanidosilver(I) diaquaargentate(I)
76. The pH of the solution containing 50 mL each of 0.10 M sodium acetate and 0.01 M acetic acid is
[Given $\mathrm{pK}_{\mathrm{a}}$ of $\mathrm{CH}_{3} \mathrm{COOH}=4.57$ ]
(1) 2.57
(2) 5.57
(3) 3.57
(4) 4.57
77. Which of the following $\mathrm{p}-\mathrm{V}$ curve represents maximum work done?
(1)

(2)

(3)

(4)

78. Match List - I with List - II.

## List-I

(Products formed)
(a) Cyanohydrin
(b) Acetal
(c) Schiff's base
(d) Oxime

Choose the correct answer from the options given below:
(1) (a)-(iv), (b)-(iii), (c)-(ii), (d)-(i)
(2) (a)-(iii), (b)-(iv), (c)-(ii), (d)-(i)
(3) (a) - (ii), (b) - (iii), (c)- (iv), (d)-(i)
(4) (a)-(i), (b)-(iii), (c)-(ii), (d)-(iv)
79. Which of the following sequence of reactions is suitable to synthesize chlorobenzene?
(1)

(2) Benzene, $\mathrm{Cl}_{2}$, anhydrous $\mathrm{FeCl}_{3}$
(3) Phenol, $\mathrm{NaNO}_{2}, \mathrm{HCl}, \mathrm{CuCl}$
(4)

80. The Kjeldahl's method for the estimation of nitrogen can be used to estimate the amount of nitrogen in which one of the following compounds?
(1)

(2)

(3)

(4)

81. Given below are two statements :

## Statement I:

The boiling points of aldehydes and ketones are higher than hydrocarbons of comparable molecular masses because of weak molecular association in aldehydes and ketones due to dipole - dipole interactions.

## Statement II:

The boiling points of aldehydes and ketones are lower than the alcohols of similar molecular masses due to the absence of H -bonding.
In the light of the above statements, choose the most appropriate answer from the options given below:
(1) Statement $I$ is incorrect but Statement II is
(2) Both Statement I and Statement II are correct.
(3) Both Statement I and Statement II are
(4) Statement I is correct but Statement II is incorrect.
82. Given below are two statements :

## Statement I:

The boiling points of the following hydrides of group 16 elements increases in the order -

$$
\mathrm{H}_{2} \mathrm{O}<\mathrm{H}_{2} \mathrm{~S}<\mathrm{H}_{2} \mathrm{Se}<\mathrm{H}_{2} \mathrm{Te} .
$$

## Statement II:

The boiling points of these hydrides increase with increase in molar mass.
In the light of the above statements, choose the most appropriate answer from the options given below:
(1) Statement I is incorrect but Statement II is correct
(2) Both Statement I and Statement II are correct
(3) Both Statement I and Statement II are incorrect
(4) Statement I is correct but Statement II is incorrect
83. In one molal solution that contains 0.5 mole of a solute, there is
(1) 1000 g of solvent
(2) 500 mL of solvent
(3) 500 g of solvent
(4) 100 mL of solvent
84. Given below are two statements : one is labelled as Assertion (A) and the other is labelled as Reason (R).

Assertion (A): ICl is more reactive than $\mathrm{I}_{2}$.
Reason (R): I-Cl bond is weaker than I-I bond.
In the light of the above statements, choose the most appropriate answer from the options given below :
(1) (A) is not correct but $(R)$ is correct.
(2) Both (A) and (R) are correct and (R) is the correct explanation of (A).
(3) Both (A) and (R) are correct but (R) is not the correct explanation of (A).
(4) (A) is correct but $(\mathrm{R})$ is not correct.
85. Gadolinium has a low value of third ionisation enthalpy because of
(1) high basic character
(2) small size
(3) high exchange enthalpy
(4) high electronegativity

## Section-B (Chemistry)

86. Compound X on reaction with $\mathrm{O}_{3}$ followed by $\mathrm{Z}_{\mathrm{n}}$ / $\mathrm{H}_{2} \mathrm{O}$ gives formaldehyde and 2-methyl propanal as
products. The compound $X$ is: products. The compound $X$ is :
(1) Pent-2-ene
(2) 3-Methylbut-1-ene
(3) 2-Methylbut-1-ene
(4) 2-Methylbut-2-ene
87. For a first order reaction $\mathrm{A} \rightarrow$ Products, initial concentration of A is 0.1 M , which becomes 0.001 M after 5 minutes. Rate constant for the reaction in
$\min ^{-1}$ is
(1) 0.2303
(2) 1.3818
(3) 0.9212
(4) 0.4606
88. In the neutral or faintly alkaline medium, $\mathrm{KMnO}_{4}$ oxidises iodide into iodate. The change in oxidation state of manganese in this reaction is from
(1) +6 to +5
(2) +7 to +4
(3) +6 to +4
(4) +7 to +3
89. A 10.0 L flask contains 64 g of oxygen at $27^{\circ} \mathrm{C}$. (Assume $\mathrm{O}_{2}$ gas is behaving ideally). The pressure inside the flask in bar is
(Given $\mathrm{R}=0.0831 \mathrm{~L} \mathrm{bar} \mathrm{K}^{-1} \mathrm{~mol}^{-1}$ )
(1) 4.9
(2) 2.5
(3) 498.6
(4) 49.8
90. Given below are two statements :

## Statement I:

In Lucas test, primary, secondary and tertiary alcohols are distinguished on the basis of their reactivity with conc. $\mathrm{HCl}+\mathrm{ZnCl}_{2}$, known as Lucas Reagent.

## Statement II:

Primary alcohols are most reactive and immediately produce turbidity at room temperature on reaction with Lucas Reagent.
In the light of the above statements, choose the most appropriate answer from the options given below:
(1) Statement I is incorrect but Statement II is correct.
(2) Both Statement I and Statement II are correct.
(3) Both Statement I and Statement II are incorrect.
(4) Statement I is correct but Statement II is incorrect.
91. $3 \mathrm{O}_{2}(\mathrm{~g}) \rightleftharpoons 2 \mathrm{O}_{3}(\mathrm{~g})$
for the above reaction at $298 \mathrm{~K}, \mathrm{~K}_{\mathrm{C}}$ is found to be $3.0 \times 10^{-59}$. If the concentration of $\mathrm{O}_{2}$ at equilibrium is 0.040 M then concentration of $\mathrm{O}_{3}$ in M is
(1) $1.2 \times 10^{21}$
(2) $4.38 \times 10^{-32}$
(3) $1.9 \times 10^{-63}$
(4) $2.4 \times 10^{31}$
92. Match List - I with List - II.

## List-I

(Ores)

## List-II

(Composition)
(a) Haematite
(b) Magnetite
(c) Calamine
(d) Kaolinite
(iv) $\left[\mathrm{Al}_{2}(\mathrm{OH})_{4} \mathrm{Si}_{2} \mathrm{O}_{5}\right]$

Choose the correct answer from the options given below:
(1) (a)-(i), (b) - (iii), (c) - (ii), (d) - (iv)
(2) (a)-(i), (b)-(ii), (c) - (iii), (d)-(iv)
(3) (a)-(iii), (b)-(i), (c) - (ii), (d)- (iv)
(4) (a)-(iii), (b)-(i), (c)-(iv), (d) - (ii)
93. Copper crystallises in fcc unit cell with cell edge length of $3.608 \times 10^{-8} \mathrm{~cm}$. The density of copper is $8.92 \mathrm{~g} \mathrm{~cm}^{-3}$. Calculate the atomic mass of copper.
(1) 65 u
(2) 63.1 u
(3) 31.55 u
(4) 60 u
94. The correctIUPAC name of the following compound is :

(1) 6-bromo-4-methyl-2-chlorohexan-4-ol
(2) 1-bromo-5-chloro-4-methylhexan-3-ol
(3) 6-bromo-2-chloro-4-methylhexan-4-ol
(4) 1-bromo-4-methyl-5-chlorohexan-3-ol
95. Find the emf of the cell in which the following reaction takes place at 298 K
$\mathrm{Ni}(\mathrm{s})+2 \mathrm{Ag}^{+}(0.001 \mathrm{M}) \rightarrow \mathrm{Ni}^{2+}(0.001 \mathrm{M})+2 \mathrm{Ag}(\mathrm{s})$
(Given that $\mathrm{E}_{\text {cell }}^{\circ}=10.5 \mathrm{~V}, \frac{2.303 \mathrm{RT}}{\mathrm{F}}=0.059$ at 298 K )
(1) 1.05 V
(2) 1.0385 V
(3) 1.385 V
(4) 0.9615 V
96. The order of energy absorbed which is responsible
(A) $\quad\left[\mathrm{Ni}\left(\mathrm{H}_{2} \mathrm{O}\right)_{2}(\mathrm{en})_{2}\right]^{2+}$
(B) $\left[\mathrm{Ni}\left(\mathrm{H}_{2} \mathrm{O}\right)_{4}(\mathrm{en})\right]^{2+}$ and
(C) $\left[\mathrm{Ni}(\mathrm{en})_{3}\right]^{2+}$
is
(1) $\quad$ (B) $>(\mathrm{A})>(\mathrm{C})$
(2) (A) $>$ (B) $>$ (C)
(3) $\quad$ (C) $>$ (B) $>$ (A)
(4) $\quad(\mathrm{C})>(\mathrm{A})>($ B $)$
97. Which one of the following is not formed when acetone reacts with 2-pentanone in the presence of dilute NaOH followed by heating?
(1)

(2)

(4)

98. If radius of second Bohr orbit of the $\mathrm{He}^{+}$ion is 105.8 pm, what is the radius of third Bohr orbit of $\mathrm{Li}^{2+}$ ion?
(1) $158.7 \AA$
(2) 158.7 pm
(3) 15.87 pm
(4) 1.587 pm
99. The pollution due to oxides of sulphur gets enhanced due to the presence of:
(a) particulate matter
(b) ozone
(c) hydrocarbons
(d) hydrogen peroxide

Choose the most appropriate answer from the options given below:
(1) (a), (c), (d) only
(2) (a), (d) only
(3) (a), (b), (d) only
(4) (b), (c), (d) only
100. The product formed from the following reaction sequence is

(i) $\mathrm{LiAlH}_{4}, \mathrm{H}_{2} \mathrm{O}$
$\xrightarrow[\text { (iii) } \mathrm{H}_{2} \mathrm{O}]{\text { (ii) } \mathrm{NaNO}_{2}+\mathrm{HCl}}$
(1)

(2)

(3)

(4)


## Section - A (Biology : Botany)

101. Which of the following is not observed during apoplastic pathway?
(1) Apoplast is continuous and does not provide any barrier to water movement.
(2) Movement of water occurs through intercellular spaces and wall of the cells.
(3) The movement does not involve crossing of cell membrane
(4) The movement is aided by cytoplasmic streaming
102. The device which can remove particulate matter present in the exhaust from a thermal power plant is :
(1) Catalytic Convertor
(2) STP
(3) Incinerator
(4) Electrostatic Precipitator
103. Which one of the following never occurs during mitotic cell division?
(1) Coiling and condensation of the chromatids
(2) Spindle fibres attach to kinetochores of chromosomes
(3) Movement of centrioles towards opposite poles
(4) Pairing of homologous chromosomes
104. Hydrocolloid carrageen is obtained from:
(1) Phaeophyceae only
(2) Chlorophyceae and Phaeophyceae
(3) Phaeophyceae and Rhodophyceae
(4) Rhodophyceae only
105. Read the following statements about the vascular bundles:
(a) In roots, xylem and phloem in a vascular bundle are arranged in an alternate manner along the different radii.
(b) Conjoint closed vascular bundles do not possess cambium
(c) In open vascular bundles, cambium is present in between xylem and phloem
(d) The vascular bundles of dicotyledonous stem possess endarch protoxylem
(e) In monocotyledonous root, usually there are more than six xylem bundles present
Choose the correct answer from the options given below:
(1) (a), (c), (d) and (e) Only
(2) (a), (b) and (d) Only
(3) (b), (c), (d) and (e) Only
(4) (a), (b), (c) and (d) Only
106. DNA polymorphism forms the basis of :
(1) Translation
(2) Genetic mapping
(3) DNA finger printing
(4) Both genetic mapping and DNA finger printing
107. Given below are two statements : one is labelled as Assertion (A) and the other is labelled as Reason (R).

## Assertion (A):

Polymerase chain reaction is used in DNA
amplification Reason (R):

The ampicillin resistant gene is used as a selectable marker to check transformation
In the light of the above statements, choose the correct answer from the options given below :
(1) (A) is not correct but ( $R$ ) is correct
(2) Both (A) and (R) are correct and correct explanation of $(\mathbf{A})$ and (R) is the
(3) Both (A) and (R) of (A)
(4) correct explanation of (A)
correct explanation of $(A)$ but $(R)$ is not the $(A)$ is correct but $(R)$ is not correct
108. The appearance of recombination nodules on homologous chromosomes during meiosis characterizes :
(1) Terminalization
(2) Synaptonemal complex
(3) Bivalent
(4) Sites at which crossing over occurs
109. The flowers are Zygomorphic in :
(a) Mustard
(b) Gulmohar
(c) Cassia
(d) Datura
(e) Chilly

Choose the correct answer from the options given below:
(c), (d), (e) Only
(a), (b), (c) Only
(b), (c) Only
(4)
(d), (e) Only
110. Production of Cucumber has increased manifold in recent years. Application of which of the following phytohormones has resulted in this increased yield as the hormone is known to produce female flowers in the plants :
(1) Cytokinin
(2) ABA
(3) Gibberellin
(4) Ethylene regarding gel electrophoresis technique?
(1) Bright orange coloured bands of DNA can be
(2) The process of extraction of separated DNA
(2) The procends from gel is called elution.
(3) The separated DNA fragments are stained by

The presence of chromogenic substrate gives
(4) The presoured DNA bands on the gel.
112. In old trees the greater pre to insect attaities and their dark brown secretion of secondary metabossels.
(b) deposition of organic compounds like tannin and resition of suberin and aro
(c) deposition of outer layer of stem.
d) deposition of tannins, gum, resial layers
(e) presence of parenchyma cells, functiona

Choose the correct answer from the options given
below:
(1)
(b) and
(d) Only
(2)
(a) and (b) Only
(3)
(c) and (d) Orily
(4)
(d) and (e) Only
113. Which of the following is incorrectly matched ?
(1) Volvox - Starch
(2) Ectocarpus - Fucoxanthin
(3) Ulothrix - Mannitol
(4) Porphyra - Floridian Starch
114. Which one of the following plants does not show plasticity?
(1) Maize
(2) Cotton
(3) Coriander
(4) Buttercup
115. Read the following statements and choose the set of correct statements :
(a) Euchromatin is loosely packed chromatin
(b) Heterochromatin is transcriptionally active
(c) Histone octomer is wrapped by negatively charged DNA in nucleosome
(d) Histones are rich in lysine and arginine
(e) A typical nucleosome contains 400 bp of DNA helix
Choose the correct answer from the options given below :
(1) (a), (c), (e) Only
(2) (b), (d), (e) Only
(3) (a), (c), (d) Only
(4) (b), (e) Only
116. Which one of the following produces nitrogen fixing nodules on the roots of Alnus?
(1) Beijernickia
(2) Rhizobium
(3) Frankia
(4) Rhodospirillum
117. Identify the incorrect statement related to Pollination:
(1) Moths and butterflies are the most dominant
(2) Polinating agents among insects

Pollination by water is quite rare in flowering
plants
(3) Pollination by wind is more common amongst
(4) Flowers produce foul odours to attract flies and beetles to get pollinated
118. Which of the following is not a method of ex situ conservation?
(1) Cryopreservation
(2) In vitro fertilization
(3) National Parks
(4) Micropropagation
119. Which one of the following is not true regarding the release of energy during ATP synthesis through chemiosmosis? It involves:
(1) Reduction of NADP to $\mathrm{NADPH}_{2}$ on the stroma side of the membrane
(2) Breakdown of proton gradient
(3) Breakdown of electron gradient
(4) Movement of protons across the membrane to the stroma
120. Which one of the following statements cannot be connected to Predation?
(1) It is necessitated by nature to maintain the ecological balance
(2) It helps in maintaining species diversity in a community
(3) It might lead to extinction of a species
(4) Both the interacting species are negatively impacted
121. Given below are two statements :

## Statement I:

Cleistogamous flowers are invariably autogamous Statement II:
Cleistogamy is disadvantageous as there is no chance for cross pollination
In the light of the above statements, choose the correct answer from the options given below:
(1) Statement I is incorrect but Statement II is correct
(2) Both Statement I and Statement II are correct
(3) Both Statement I and Statement II are incorrect
(4) Statement I is correct but Statement II is incorrect
122. Given below are two statements :

## Statement I:

The primary $\mathrm{CO}_{2}$ acceptor in $\mathrm{C}_{4}$ plants is phosphoenolpyruvate and is found in the mesophyll cells.

## Statement II:

Mesophyll cells of $\mathrm{C}_{4}$ plants lack RuBisCo enzyme. In the light of the above statements, choose the correct answer from the options given below :
(1) Statement $I$ is incorrect but Statement II is correct
(2) Both Statement I and Statement II are correct
(3) Both Statement I and Statement II are
(4) Statement I is correct but Statement II is
incorrect
123. Identify the correct set of statements:
(a) The leaflets are modified into pointed hard thorns in Citrus and Bougainvillea
(b) Axillary buds form slender and spirally coiled
(c) Stem is flacumber and pumpkin
modified to perform the function of leaves
(d) Rhizophora shows vertically upward growing roots that help to get oxygen for respiration
(e) Subaerially growing stems in grasses and strawberry help in vegetative propagation Choose the correct answer from the options given
below:
(1) (a), (b), (d) and (e) Only
(2) (b) and (c) Only
(3) (a) and (d) Only
(4) (b), (c), (d) and (e) Only
124. The gaseous plant growth regulator is used in plants to:
(1) kill dicotyledonous weeds in the fields
(2) speed up the malting process
(3) promote root growth and roothair formation to increase the absorption surface
(4) help overcome apical dominance
125. Which one of the following plants shows vexillary aestivation and diadelphous stamens ?
(1) Solanum nigrum
(2) Colchicum autumnale
(3) Pisum sativum
(4) Allium cepa
126. XO type of sex determination can be found in:
(1) Monkeys
(2) Drosophila
(3) Birds
(4) Grasshoppers
127. What amount of energy is released from glucose during lactic acid fermentation?
(1) Less than 7\%
(2) Approximately 15\%
(3) More than $18 \%$
(4) About $10 \%$
128. Given below are two statements :

## Statement I:

Decomposition is a process in which the detritus is degraded into simpler substances by microbes.
Statement II :
Decomposition is faster if the detritus is rich in lignin and chitin
In the light of the above statements, choose the correct answer from the options given below :
(1) Statement I is incorrect but Statement II is correct
(2) Both Statement I and Statement II are correct
(3) Both Statement I and Statement II are
(4) Statement I is correct but Statement II is incorrect
129. What is the net gain of ATP when each molecule of glucose is converted to two molecules of pyruvic acid ?
(1) Eight
(2) Four
(3) Six
(4) Two
130. Match List-I with List - II

## List-I

(a) Manganese
(b) Magnesium
(c) Boron
(d) Iron

List-II
(i) Activates the enzyme
(ii) Required for pollen germination
(iii) Activates enzymes of
respiration
(iv) Functions in splitting of
water during photosynthesis

Choose the correct answer from the options given below:
(a) - (iii), (b) - (i), (c) - (ii), (d) - (iv)
(2) (a)-(iii), (b) - (iv), (c)- (i), (d) - (ii)
(3)
(a) - (iv), (b) - (iii), (c) - (ii), (d) - (i)
(4)
(a) - (iv), (b) - (i), (c) - (ii), (d) - (iii)
131. Habitat loss and fragmentation, over exploitation, alien species invasion and co-extinction are causes for:
(1) Natality
(2) Population explosion
(3) Competition
(4) Biodiversity loss
132. The process of translation of mRNA to proteins begins as soon as :
(1) The tRNA is activated and the larger subunit of ribosome encounters mRNA
(2) The small subunit of ribosome encounters mRNA
(3) The larger subunit of ribosome encounters
(4) Both the subunits join together to bind with
133. "Girdling Experiment" was performed by Plant Physiologists to identify the plant tissue through which:
(1) osmosis is observed
(2) water is transported
(3) food is transported
(3) food is transported
(4) for both water and food transportation
134. Exoskeleton of arthropods is composed of:
(1) Glucosamine
(2) Cutin
(3) Cellulose
(4) Chitin
135. Given below are two statements :

## Statement I:

Mendel studied seven pairs of contrasting traits in pea plants and proposed the Laws of Inheritance Statement II:

Seven characters examined by Mendel in his experiment on pea plants were seed shape and position and stem height
In the light of the above statements, choose the correct answer from the options given below :
(1) Statement I is incorrect but Statement II is
(2) Both Statement I and Statement II are correct
(3) Both Statement I and Statement II are incorrect
(4) Statement I is correct but Statement II is

## Section - B (Biology : Botany)

136. Which part of the fruit, labelled in the given figure makes it a false fruit?

(1) $D \rightarrow$ Seed
(2) $\mathrm{A} \rightarrow$ Mesocarp
(3) $\mathrm{B} \rightarrow$ Endocarp
(4) $\mathrm{C} \rightarrow$ Thalamus
137. Addition of more solutes in a given solution will :
(1) not affect the water potential at all
(2) raise its water potential
(3) lower its water potential
(4) make its water potential zero
138. The anatomy of springwood shows some peculiar springwood.
(a) It is also called as the earlywood
(b) In spring season cambium produces xylem elements with narrow vessels
(c) It is lighter in colour
(d) The springwood along with autumnwood shows alternate concentric rings forming annual rings
(e) It has lower density

Choose the correct answer from the options given
below:
(1)
(c), (d) and (e) Only
(2) (a), (b), (d) and (e) Only
(3) (a), (c), (d) and (e) Only
(4) (a), (b) and (d) Only
139. Which one of the following will accelerate phosphorus cycle?
(1) Rain fall and storms
(2) Burning of fossil fuels
(3) Volcanic activity
(4) Weathering of rocks
140. The entire fleet of buses in Delhi were converted to CNG from diesel. In reference to this, which one of the following statements is false?
(1) It can not be adulterated like diesel
(2) CNG burns more efficiently than diesel
(3) The same diesel engine is used in CNG buses making the cost of conversion low
(4) It is cheaper than diesel
141. What is the role of large bundle shealth cells found around the vascular bundles in $\mathrm{C}_{4}$ plants ?
(1) To protect the vascular tissue from high light
intensity
(2) To provide the site for photorespiratory
pathway
(3) To increase the number of chloroplast for the
(4) To enable the plant to tolerate high
temperature
142. Which of the following occurs due to the presence of autosome linked dominant trait?
(1) Thalessemia
(2) Sickle cell anaemia
(3) Myotonic dystrophy
(4) Haemophilia
143. Read the following statements on lipids and find out correct set of statements :
(a) Lecithin found in the plasma membrane is a glycolipid
(b) Saturated fatty acids possess one or more $\mathrm{c}=\mathrm{c}$ bonds
(c) Gingely oil has lower melting point, hence remains as oil in winter
(d) Lipids are generally insoluble in water but soluble in some organic solvents
(e) When fatty acid is esterified with glycerol, monoglycerides are formed
Choose the correct answer from the options given below :
(1) (a), (b) and (d) only
(2) (a), (b) and (c) only
(3) (a), (d) and (e) only
(4) (c), (d) and (e) only
144. If a geneticist uses the blind approach for sequencing the whole genome of an organism, followed by assignment of function to different segments, the methodology adopted by him is called as :
(1) Bioinformatics
(2) Sequence annotation
(3) Gene mapping
(4) Expressed sequence tags
145. In the following palindromic base sequences of DNA, which one can be cut easily by particular restriction enzyme?
(1) $5^{\prime}$ GTATTC $3^{\prime} ; 3^{\prime}$ CATAAG5 ${ }^{\prime}$
(2) $5^{\prime}$ GATACT $3^{\prime} ; 3^{\prime}$ СTATGA $5^{\prime}$
(3) $5^{\prime}$ GAATTC3'; $3^{\prime}$ CTTAAG5 ${ }^{\prime}$
(4) $5^{\prime}$ СТСАGT3'; $3^{\prime}$ GAGTCA $5^{\prime}$
146. Match the plant with the kind of life cycle it exhibits :

## List - I

(a) Spirogyra
(b) Fern
(c) Funaria
(d) Cycas

## List-II

(i) Dominant diploid sporophyte vascular plant, with highly reduced male or female gametophyte
(ii) Dominant haploid free-living gametophyte
(iii) Dominant diploid sporophyte alternating with reduced gametophyte called prothallus
(iv) Dominant haploid leafy gametophyte alternating with partially dependent multicellular sporophyte

Choose the correct answer from the options given below:
(a) - (ii), (b) - (iv), (c) - (i), (d) - (iii)
(2) (a)-(iv), (b)-(i), (c)-(ii), (d)- (iii)
(3)
(a)-(ii), (b) - (iii), (c)-(iv), (d) - (i)
(4)
(a) - (iii), (b) - (iv), (c) - (i), (d) - (ii)
147. While explaining interspecific interaction of population, $(+)$ sign is assigned for beneficial interaction, ( - ) sign is assigned for detrimental interaction and ( 0 ) for neutral interaction. Which of the following interactions can be assigned ( + ) for one species and ( - ) for another species involved in the interaction?
(1) Competition
(2) Predation
(3) Amensalism
(4) Commensalism
148. Transposons can be used during which one of the following?

## (1) Gene sequencing

(2) Polymerase Chain Reaction
(3) Gene silencing
(4) Autoradiography
149. Match List - I with List - II.

List-I
(a) Metacentric chromosome
(b) Acrocentric chromosome
(c) Submetacentric
(d) Telocentric chromosome

## List - II

(i) Centromere situated close to the end forming one extremely short and one very long arms
(ii) Centromere at the terminal end
(iii) Centromere in the middle forming two equal arms of chromosomes
(iv) Centromere slightly away from the middle forming one shorter arm and one longer arm

Choose the correct answer from the options given below:
(1) (a)-(i), (b) - (ii), (c) - (iii), (d) - (iv)
(2) (a)-(iii), (b)-(i), (c)-(iv), (d)-(ii)
(a) - (i), (b) - (iii), (c) - (ii), (d) - (iv)
(4) (a)-(ii), (b)-(iii), (c)-(iv), (d)-(i)
150. Given below are two statements : one is labelled as Assertion (A) and the other is labelled as Reason (R).

## Assertion (A):

Mendel's law of Independent assortment does not hold good for the genes that are located closely on the same chromosome.
Reason (R):
Closely located genes assort independently.
In the light of the above statements, choose the correct answer from the options given below :
(1) (A) is not correct but ( $R$ ) is correct
(2) Both (A) and (R) are correct and (R) is the correct explanation of (A)
(3) Both (A) and (R) are correct but (R) is not the correct explanation of (A)
(4) (A) is correct but ( $R$ ) is not correct

## Section - A (Biology : Zoology)

151. Which of the following statements with respect to Endoplasmic Reticulum is incorrect?
(1) SER are the sites for lipid synthesis
(2) RER has ribosomes attached to ER
(3) SER is devoid of ribosomes
(4) In prokaryotes only RER are present
152. Regarding Meiosis, which of the statements is incorrect?
(1) Four haploid cells are formed at the end of Meiosis-II
(2) There are two stages in Meiosis, Meiosis-I and II
(3) DNA replication occurs in $S$ phase of Meiosis-II
(4) Pairing of homologous chromosomes and recombination occurs in Meiosis-I
153. Given below are two statements :

## Statement I:

The coagulum is formed of network of threads called thrombins.

## Statement II:

Spleen is the graveyard of erythrocytes.
In the light of the above statements, choose the most appropriate answer from the options given below:
(1) Statement I is incorrect but Statement II is correct
(2) Both Statement I and Statement II are correct
(3) Both Statement I and Statement II are incorrect
(4) Statement I is correct but Statement II is incorrect
154. Given below are two statements :

## Statement I:

Restriction endonucleases recognise specific sequence to cut DNA known as palindromic nucleotide sequence.

## Statement II :

Restrictionendonucleases cut the DNA strand a little away from the centre of the palindromic site.
In the light of the above statements, choose the most appropriate answer from the options given below :
(1) Statement I is incorrect but Statement II is
correct
(2) Both Statement I and Statement II are correct
(3) Both Statement I and Statement II are
incorrect
(4) Statement I is correct but Statement II is
incorrect
155. Identify the asexual reproductive structure associated with Penicillium :
(1) Buds
(2) Zoospores
(3) Conidia
(4) Gemmules
156. Nitrogenous waste is excreted in the form of pellet or paste by :
(1) Pavo
(2) Ornithorhynchus
(3) Salamandra
(4) Hippocampus
157. Which of the following is present between the adjacent bones of the vertebral column?
(1) Smooth muscle
(2) Intercalated discs
(3) Cartilage
(4) Areolar tissue
158. Given below are two statements : one is labelled as Assertion (A) and the other is labelled as Reason (R).

## Assertion (A):

All vertebrates are chordates but all chordates are not vertebrates.
Reason (R) :
Notochord is replaced by vertebral column in the adult vertebrates.
In the light of the above statements, choose the most appropriate answer from the options given below :
(1) (A) is not correct but ( $R$ ) is correct
(2) Both (A) and (R) are correct and (R) is the correct explanation of (A)
(3) Both (A) and (R) are correct but (R) is not the correct explanation of (A)
(4) (A) is correct but ( $R$ ) is not correct
159. Lippe's loop is a type of contraceptive used as :
(1) Copper releasing IUD
(2) Cervical barrier
(3) Vault barrier
(4) Non-Medicated IUD
160. At which stage of life the oogenesis process is initiated?
(1) Adult
(2) Puberty
(3) Embryonic development stage
(4) Birth
161. Which of the following is not a connective tissue?
(1) Neuroglia
(2) Blood
(3) Adipose tissue
(4) Cartilage
162. Given below are two statements : one is labelled as Assertion (A) and the other is labelled as Reason (R).

## Assertion (A) :

Osteoporosis is characterised by decreased bone mass and increased chances of fractures.
Reason (R) :
Common cause of osteoporosis is increased levels of estrogen.
In the light of the above statements, choose the most appropriate answer from the options given below :
(1) (A) is not correct but ( $R$ ) is correct
(2) Both (A) and (R) are correct and (R) is the correct explanation of (A)
(3) Both (A) and (R) are correct but (R) is not the correct explanation of (A)
(4) (A) is correct but ( $R$ ) is not correct
163. Which of the following statements are true for spermatogenesis but do not hold true for Oogenesis?
(a) It results in the formation of haploid gametes
(b) Differentiation of gamete occurs after the completion of meiosis
(c) Meiosis occurs continuously in a mitotically dividing stem cell population
(d) It is controlled by the Luteinising hormone (LH) and FollicleStimulating Hormone (FSH) secreted by the anterior pituitary
(e) It is initiated at puberty

Choose the most appropriate answer from the options given below :
(1) (b), (c) and (e) only
(2) (c) and (e) only
(3) (b) and (c) only
(4) (b), (d) and (e) only
164. A dehydration reaction links two glucose molecules to produce maltose. If the formula for glucose is $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{6}$ then what is the formula for maltose?
(1) $\mathrm{C}_{12} \mathrm{H}_{24} \mathrm{O}_{11}$
(2) $\mathrm{C}_{12} \mathrm{H}_{20} \mathrm{O}_{10}$
(3) $\mathrm{C}_{12} \mathrm{H}_{24} \mathrm{O}_{12}$
(4) $\mathrm{C}_{12} \mathrm{H}_{22} \mathrm{O}_{11}$
165. Breeding crops with higher levels of vitamins and
minerals or higher proteins and healthier fals minerals or higher proteins and healthier fats is
(1) Bio-accumulation
(2) Bio-magnification
(3) Bio-remediation
(4) Bio-fortification
166. Identify the microorganism which is responsible for the production of an immunosuppressive molecule cyclosporin A :
(1) Streptococcus cerevisiae
(2) Trichoderma polysporum
(3) Clostridium butylicum
(4) Aspergillus niger
167. Tegmina in cockroach, arises from :
(1) Prothorax and Mesothorax
(2) Prothorax
(3) Mesothorax
(4) Metathorax
168. Detritivores breakdown detritus into smaller particles. This process is called :
(1) Decomposition
(2) Catabolism
(3) Fragmentation
(4) Humification
169. If ' 8 ' Drosophila in a laboratory population of ' 80 ' died during a week, the death rate in the population is $\qquad$ individuals per Drosophila per week
(1) zero
(2) 0.1
(3) 10
(4) 1.0
170. In which of the following animals, digestive tract has additional chambers like crop and gizzard ?
(1) Pavo, Psittacula, Corvus
(2) Corvus, Columba, Chameleon
(3) Bufo, Balaenoptera, Bangarus
(4) Catla, Columba, Crocodilus
171. Given below are two statements :

Statement I:
The release of sperms into the seminiferous tubules is called spermiation.
Statement II:
Spermiogenesis is the process of formation of sperms from spermatogonia.
In the light of the above statements, choose the most appropriate answer from the options given below :
(1) Statement I is incorrect but Statement II is
(2) Both Statement I and Statement II are correct
(3) Both Statement I and Statement II are
(4) Statement I is correct but Statement II is incorrect
172. Which of the following functions is not performed by secretions from salivary glands?
(1) Digestion of disaccharides
(2) Control bacterial population in mouth
(3) Digestion of complex carbohydrates
(4) Lubrication of oral cavity
173. In-situ conservation refers to :
(1) Conserve only extinct species
(2) Protect and conserve the whole ecosystem
(3) Conserve only high risk species
(4) Conserve only endangered species
174. In gene therapy of Adenosine Deaminase (ADA) deficiency, the patient requires periodic infusion of genetically engineered lymphocytes because:
(1) Genetically engineered lymphocytes are not immortal cells.
(2) Retroviral vector is introduced into these lymphocytes.
(3) Gene isolated from marrow cells producing ADA is introduced into cells at embryonic stages
(4) Lymphocytes from patient's blood are grown in culture, outside the body.
175. Given below are two statements :

## Statement I:

Mycoplasma can pass through less than 1 micron filter size.

## Statement II :

Mycoplasma are bacteria with cell wall
In the light of the above statements, choose the most appropriate answer from the options given below:
(1) Statement I is incorrect but Statement II is correct
(2) Both Statement I and Statement II are correct
(3) Both Statement I and Statement II are incorrect
(4) Statement I is correct but Statement II is incorrect
176. Which of the following is not the function of conducting part of respiratory system ?
(1) Provides surface for diffusion of $\mathrm{O}_{2}$ and $\mathrm{CO}_{2}$
(2) It clears inhaled air from foreign particles
(3) Inhaled air is humidified
(4) Temperature of inhaled air is brought to body
temperature
177. In the taxonomic categories which hierarchial arrangement in ascending order is correct in case of
animals?
(1) Kingdom, Order, Phylum, Class, Family,
(2) Kingdom, Phylum, Class, Order, Family,
(3) Kingdom, Class, Phylum, Family, Order,
Genus, Species
(4) Kingdom, Order, Class, Phylum, Family,
Genus, Species
178. Given below are two statements :

Statement I:
Fatty acids and glycerols cannot be absorbed into
the blood.

## Statement II :

Specialized lymphatic capillaries called lacteals carry chylomicrons into lymphatic vessels and ultimately into the blood.
In the light of the above statements, choose the most
appropriate answer from the options given below :
(1) Stat
(1) Statement I is incorrect but Statement II is
correct
(2) Both Statement I and Statement II are correct
(3) Both Statement I and Statement II are correct incorrect
(4) Statement I is correct but Statement II is
incorrect
179. Given below are two statements :

Statement I:
Autoimmune disorder is a condition where body
defense mechanism recognizes its own cells as foreign bodies.
Statement II:
Rheumatoid art
not attack self cells.
In the light of the above statements,
(1) appropriate answer from the
(1) Statement I is incor the options given below:
(2) Correct incorrect but Statement II is
(3) Both Statement I and Statement II are correct
(4) incorrect Statement I is correct but Statement II is
incorrect
180. In an E.coli strain $i$ gene gets mutated and its product can not bind the inducer molecule. If growth medium is provided with lactose, what will be the outcome?
(1) RNA polymerase will bind the promoter
(2) Only $z$ gene will get transcribed
(3) $z, y$, a genes will be transcribed
(4) $z, y$, a genes will not be translated
181. Select the incorrect statement with reference to mitosis:
(1) Splitting of centromere occurs at anaphase.
(2) All the chromosomes lie at the equator at
(3) Spindle fibres attach to centromere of
(4) Chromosomes decondense at telophase.
182. Which of the following is a correct match for disease
and its symptoms ?
(1) Muscular dystrophy - An auto immune disorder causing progressive degeneration of
(2) Arthritis - Inflammed joints
(3) Tetany - high $\mathrm{Ca}^{2+}$ level causing rapid
(4) Myasthenia gravis - Genetic disorder resulting in weakening and paralysis of
skeletal muscle
183. Natural selection where more individuals acquire specific character value other than the mean character value, leads to:
(1) Random change
(2) Stabilising change
(3) Directional change
(4) Disruptive change
184. Under normal physiological conditions in human being every 100 ml of oxygenated blood can deliver (1) 10 ml
(2) 2 ml
(3) 5 ml
(4) 4 ml
185. If the length of a DNA molecule is 1.1 metres, what will be the approximate number of base pairs?
(1)
(2) $3.3 \times 10^{9} \mathrm{bp}$
(3) $6.6 \times 10^{9} \mathrm{bp}$
(4) $3.3 \times 10^{6} \mathrm{bp}$
186. Select the incorrect statement with respect to acquired immunity.
(1) Acquired immunity is non-specific type of
(2) Primary response is produced when our body
(3) Anamnestic response is elicited on subsequent encounters with the same
pathogen.
(4) Anamnestic response is due to memory of first encounter.

## 187. Match List-I with List - II.

## List-I

(a) Bronchioles
(b) Goblet cell
(c) Tendons
(d) Adipose Tissue

## List-II

(i) Dense Regular Connective Tissue
(ii) Loose Connective Tissue
(iii) Glandular Tissue
(iv) Ciliated Epithelium

Choose the correct answer from the options given
below:
(1) (a)-(iii), (b) - (iv), (c) - (ii), (d) - (i)
(2) (a)-(iv), (b) - (iii), (c) -(i), (d) - (ii)
(4) (a)-(ii), (b)-(i), (c) - (iv), (d)-(iii)
188. The recombination frequency between the genes a \& c is $5 \%, \mathrm{~b} \& \mathrm{c}$ is $15 \%, \mathrm{~b} \& \mathrm{~d}$ is $9 \%, \mathrm{a} \& \mathrm{~b}$ is $20 \%$, c \& d is $24 \%$ and a \& $d$ is $29 \%$. What will be the sequence of these genes on a linear chromosome?
(1) $a, c, b, d$
(2) $a, d, b, c$
(3) $d, b, a, c$
(4) $a, b, c, d$
189. Which one of the following statements is correct?
(1) Increased ventricular pressure causes closing of the semilunar valves.
(2) The atrio-ventricular node (AVN) generates an action potential to stimulate atrial contraction
(3) The tricuspid and the bicuspid valves open due to the pressure exerted by the simultaneous contraction of the atria
(4) Blood moves freely from atrium to the ventricle during.joint diastole.
190. Statements related to human Insulin are given below. Which statement(s) is/are correct about genetically engineered Insulin?
(a) Pro-hormone insulin contain extra stretch of C-peptide
(b) A-peptide and B-peptide chains of insulin were produced separately in E.coli, extracted and combined by creating disulphide bond between them.
(c) Insulin used for treating Diabetes was extracted from Cattles and Pigs.
(d) Pro-hormone Insulin needs to be processed for converting into a mature and functional hormone.
(e) Some patients develop allergic reactions to the foreign insulin.
Choose the most appropriate answer from the options given below :
(1) (c), (d) and (e) only
(2) (a), (b) and (d) only
(3) (b) only
(4)
(c) and (d) only
191. If a colour blind female marries a man whose mother was also colour blind, what are the chances of her progeny having colour blindness?
(1) $100 \%$
(2) $25 \%$
(3) $50 \%$
(4) $75 \%$
192. Which of the following are not the effects of Parathyroid hormone?
(a) Stimulates the process of bone resorption
(b) Decreases $\mathrm{Ca}^{2+}$ level in blood
(c) Reabsorption of $\mathrm{Ca}^{2+}$ by renal tubules
(d) Decreases the absorption of $\mathrm{Ca}^{2+}$ from digested food
(e) Increases metabolism of carbohydrates

Choose the most appropriate answer from the
options given below: options given below:
(1) (b) and (c) only
(2) (a) and (c) only
(3) (b), (d) and (e) only
(4) (a) and (e) only
193. Ten E.coli cells with ${ }^{15} \mathrm{~N}$-dsDNA are incubated in medium containing ${ }^{14} \mathrm{~N}$ nucleotide. After 60 minutes, how many E.coli cells will have DNA totally
free from ${ }^{15} \mathrm{~N}$ ?
(1) 80 cells
(2) 20 cells
(3) 40 cells
(4) 60 cells

Select the incorrect statement regarding synapses: 25
(1) Impulse tran
(1) Impulse transmissiont regarding synapses: synapse is always fasten across a chemical electrical synapse.
(2) The membranes of postsynaptic neurons of presynaptic and in an electrical synapse. in close proximity Electrical current can flow directly from one neuron into the other across the electrical
synapse. Chemical synapses use neurotransmitters
195. Given below are two statements :

Statement I:
In a scrubber the exhaust from the thermal plant is passed through the electric wires to charge the dust
particles.

## Statement II:

Particulate matter (PM 2.5) can not be removed by scrubber but can be removed by an electrostatic
precipitator.
In the light of the above statements, choose the most appropriate answer from the options given below :
(1) Statement I is incorrect but Statement II is
(2) Both Statement I and Statement II are correct
(3) Both Statement I and Statement II are incorrect
(4) Statement I is correct but Statement II is incorrect
196. Which of the following is not a desirable feature of a cloning vector?
(1) Presence of two or more recognition sites
(2) Presence of origin of replication
(3) Presence of a marker gene
(4) Presence of single restriction enzyme site
197. Which of the following statements is not true?
(1) Flippers of penguins and dolphins are a pair of homologous organs
(2) Analogous structures are a result of convergent evolution
(3) Sweet potato and potato is an example of analogy
(4) Homology indicates common ancestry
198. Match List - I with List - II with respect to methods of Contraception and their respective actions.

## List-I

(a) Diaphragms
(b) Contraceptive Pills

List-II
(i) Inhibit ovulation and Implantation
(c) Intra Uterine
(d) Lactational Amenorrhea
(ii) Increase phagocytosis of sperm within Uterus
(iii) Absence of Menstrual cycle and ovulation following parturition
(iv) They cover the cervix blocking the entry of
sperms

Choose the correct answer from the options given below:
(1) (a) - (iii), (b) - (ii), (c)- (i), (d) -(iv)
(2) (a)-(iv), (b)-(i), (c)-(iii), (d)-(ii)
(3) ' (a) -(iv), (b)-(i), (c) - (ii), (d) - (iii)
(4) (a)-(ii), (b)-(iv), (c)-(i), (d) -(iii)
199. Match List - I with List - II.

List -I
(Biological Molecules)

| (a) | Glycogen | (i) |
| :--- | :--- | :--- |
| (b) Hormone |  |  |
| (b) Globulin | (ii) | Biocatalyst |
| (c) | Steroids | (iii) |
| Antibody |  |  |
| (d) | Thrombin | (iv) |
| Choose the correct answe product |  |  |

## List-II

(Biological functions)
(i) Hormone
(ii) Biocatalyst
(iii) Antibody
(iv) Storage product

Choose the correct answer from the options given below:
(a)-(iv), (b)-(iii), (c)-(i), (d)- (ii)
(a)-(iii), (b)-(ii), (c)-(iv), (d)- (i)
(a)-(iv), (b)-(ii), (c)-(i), (d)-(iii)
(a)-(ii), (b)-(iv), (c)-(iii), (d)-(i)
200. Which of the following is a correct statement?
(1) Mycoplasma have DNA, Ribosome and cell wall
(2) Cyanobacteria are a group of autotrophic organisms classified under Kingdom Monera.
(3) Bacteria are exclusively heterotrophic organisms.
(4) Slime moulds are saprophytic organisms classified under Kingdom Monera.

