## **PHYSICS**

### **SECTION-A**

- 1. Given below are two statement: one is labelled as Assertion A and the other is labelled as Reason R.
  - Assertion A : When a rod lying freely is heated, no thermal stress is developed in it.
  - Reason R: On heating the length of the rod increases.

In the light of the above statements, choose the correct answer from the options given below:

- (1) Both A and R are true but R is NOT the correct explanation of A
- (2) A is false but R is true
- (3) A is true but R is false
- (4) Both A and R are true and R is the correct explanation of A

## Official Ans. by NTA (1)

- **Sol.** A and R are true but R is not the correct explanation of A.
- 2. A student is performing the experiment of resonance column. The diameter of the column tube is 6 cm. The frequency of the tuning fork is 504 Hz. Speed of the sound at the given temperature is 336 m/s. The zero of the meter scale coincides with the top end of the resonance column tube. The reading of the water level in the column when the first resonance occurs is:
  - (1) 13 cm
- (2) 16.6 cm
- (3) 18.4 cm
- (4) 14.8 cm

### Official Ans. by NTA (4)

**Sol.** d = 6cm, f = 504, v = 336 m/s e = 0.3d

$$l + e = \frac{\lambda}{4} = \frac{V}{4f}$$

$$l = 16.66 - 0.3 \times 6$$

l = 14.866 cm

l = 14.8 cm

3. Two satellites A and B of masses 200kg and 400kg are revolving round the earth at height of 600 km and 1600 km respectively. If  $T_A$  and  $T_B$  are the time periods of A and B respectively then the value of  $T_B - T_A$ :



[Given : radius of earth = 6400km, mass of earth =  $6 \times 10^{24}$  kg]

- (1)  $1.33 \times 10^3$  s
- $(2) 3.33 \times 10^2 \text{ s}$
- $(3) 4.24 \times 10^3 \text{ s}$
- $(4) 4.24 \times 10^2 \text{ s}$

Official Ans. by NTA (1)

**Sol.** 
$$T = 2\pi \sqrt{\frac{r^3}{GM}}$$

$$T_{A} = 2\pi \sqrt{\frac{(6400 + 600) \times 10^{3}}{GM}}$$

$$T_A = 2\pi \times 10^9 \sqrt{\frac{7^3}{GM}}$$

$$T_{\rm B} = 2\pi \times 10^9 \sqrt{\frac{8^3}{\rm GM}}$$

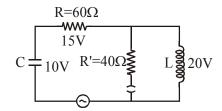
$$T_B - T_A = \frac{2\pi 10^9}{\sqrt{GM}} \left[ 8\sqrt{8} - 7\sqrt{7} \right]$$

$$= 314 \times 4.107$$

$$= 1289.64$$

$$= 1.289 \times 10^3 \text{ s}$$

4. The angular frequency of alternating current in a L-C-R circuit is 100 rad/s. The components connected are shown in the figure. Find the value of inductance of the coil and capacity of condenser.



- (1) 0.8 H and 150  $\mu F$
- (2) 0.8 H and 250 μF
- (3) 1.33 H and 250 μF
- (4) 1.33 H and 150 μF

Official Ans. by NTA (2)

**Sol.** Current through  $60\Omega$  resistance  $=\frac{15}{60} = \frac{1}{4}$  A

thus capacitor current =  $\frac{1}{4}$  A

$$V_C = I X_C$$

$$10 = \frac{1}{4} \times \frac{1}{\omega C}$$

$$\therefore C = \frac{1}{40\omega} = \frac{1}{4000} = 250 \ \mu F$$

Now,

current through  $40\Omega$  resistance =  $\frac{20}{40} = \frac{1}{2}$  A

thus current through inductor  $=\frac{1}{2} - \frac{1}{4} = \frac{1}{4}$  A

$$V_L = IX_L = \frac{1}{4} \times \omega L$$

$$20 = \frac{1}{4} \times 100 \times L$$

$$\Rightarrow$$
 L = 0.8 H

5. A proton, a deuteron and an α particle are moving with same momentum in a uniform magnetic field. The ratio of magnetic forces acting on them is \_\_\_\_ and their speed is \_\_\_\_ in the ratio.

(1) 1:2:4 and 2:1:1

(2) 2:1:1 and 4:2:1

(3) 4:2:1 and 2:1:1

(4) 1:2:4 and 1:1:2

Official Ans. by NTA (2)

**Sol.** 
$$F = q(\vec{v} \times \vec{B}) = \frac{q}{m}(\vec{P} \times \vec{B})$$

$$\Rightarrow F \propto \frac{q}{m}$$

thus 
$$F_1: F_2: F_3 = \frac{q_1}{m_1}: \frac{q_2}{m_2}: \frac{q_3}{m_3}$$

$$= \frac{e}{m_p} : \frac{e}{2m_p} : \frac{2e}{4m_p}$$

$$= \frac{1}{1} : \frac{1}{2} : \frac{2}{4}$$

$$= 2 : 1 : 1$$

Now for speed calculation

$$P = constant \implies v \propto \frac{1}{m}$$

thus 
$$v_1 : v_2 : v_3 = \frac{1}{m_p} : \frac{1}{2m_p} : \frac{1}{4m_p}$$

$$=\frac{1}{1}:\frac{1}{2}:\frac{1}{4}$$

- **6.** Given below are two statement :
  - Statement-I: A speech signal of 2 kHz is used to modulate a carrier signal of 1 MHz. The band width requirement for the signal is 4 kHz.

Statement-II: The side band frequencies are 1002 kHz. and 998 kHz.

In the light of the above statements, choose the correct answer from the options given below:

- (1) Statement I is true but Statement II is false
- (2) Statement I is false but Statement II is true
- (3) Both Statement I and Statement II are true
- (4) Both Statement I and Statement II are false

### Official Ans. by NTA (3)

**Sol.**  $f_m = 2kHz$ 

$$f_c = 1MHz = 1000 \text{ kHz}$$

Band width =  $2f_m = 4kHz$ 

.. Side frequencies will be

- $= f_c \pm f_m$
- $= (1000 \pm 2) \text{ kHz}$
- = 998 kHz & 1002 kHz

So statement-I & statement-II both are correct.

- 7. If the time period of a two meter long simple pendulum is 2s, the acceleration due to gravity at the place where pendulum is executing S.H.M. is:
  - (1)  $\pi^2 \text{ms}^{-2}$
- $(2) 9.8 \text{ ms}^{-2}$
- (3)  $2\pi^2 \text{ms}^{-2}$
- $(4) 16 \text{ m/s}^2$

## Official Ans. by NTA (3)

**Sol.** 
$$T = 2\pi \sqrt{\frac{l}{g}}$$

$$2 = 2\pi \sqrt{\frac{2}{g}}$$

$$\Rightarrow$$
 g =  $2\pi^2$ 

- 8. The pitch of the screw gauge is 1mm and there are 100 divisions on the circular scale. When nothing is put in between the jaws, the zero of the circular scale lies 8 divisions below the reference line. When a wire is placed between the jaws, the first linear scale division is clearly visible while 72nd division on circular scale coincides with the reference line. The radius of the wire is
  - (1) 1.64 mm
- (2) 0.82 mm
- (3) 1.80 mm
- (4) 0.90 mm

Official Ans. by NTA (2)

**Sol.** Least count = 
$$\frac{1 \text{mm}}{100}$$
 = 0.01mm

zero error = 
$$+ 8 \times LC = + 0.08$$
 mm  
True reading (Diameter)

$$= (1 \text{ mm} + 72 \times \text{LC}) - (\text{Zero error})$$

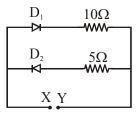
$$= (1mm + 72 \times 0.01 mm) - 0.08 mm$$

$$= 1.72 mm - 0.08 mm$$

$$= 1.64 mm$$

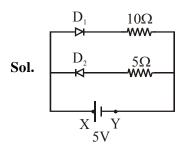
therefore, radius = 
$$\frac{1.64}{2}$$
 = 0.82 mm.

9. A 5V battery is connected across the points X and Y. Assume D<sub>1</sub> and D<sub>2</sub> to be normal silicon diodes. Find the current supplied by the battery if the +ve terminal of the battery is connected to point X.



- $(1) \sim 0.5 \text{ A}$
- $(2) \sim 1.5 \text{ A}$
- $(3) \sim 0.86 \text{ A}$
- $(4) \sim 0.43 \text{ A}$

Official Ans. by NTA (4)



Here only  $D_1$  will work and we know for silicon diode, potential drop on  $D_1$  will be  $0.7\mathrm{V}$ 

$$I = \frac{5 - 0.7}{10} = 0.43 \text{ A}$$

10. An  $\alpha$  particle and a proton are accelerated from rest by a potential difference of 200 V. After this, their de Broglie wavelengths are  $\lambda_{\alpha}$  and

 $\lambda_p$  respectively. The ratio  $\frac{\lambda_p}{\lambda_\alpha}$  is :

- (1) 3.8
- (2) 8
- (3) 7.8
- (4) 2.8

Official Ans. by NTA (4)

$$Sol. \quad \lambda = \frac{h}{p} = \frac{h}{\sqrt{2m\,qV}}$$

$$\frac{\lambda_{p}}{\lambda_{\alpha}} = \sqrt{\frac{m_{\alpha} q_{\alpha}}{m_{p} q_{p}}} = \sqrt{\frac{4m_{p} \times 2e}{m_{p} \times e}} = \sqrt{8} = 2\sqrt{2}$$

$$= 2\sqrt{2}$$

$$\frac{\lambda_p}{\lambda_q} = 2 \times 1.4 = 2.8$$

11. A diatomic gas, having  $C_p = \frac{7}{2}R$  and

 $C_v = \frac{5}{2}R$ , is heated at constant pressure. The

ratio dU: dQ: dW:

- (1) 5:7:3
- (2) 5:7:2
- $(3) \ 3:7:2$
- $(4) \ 3:5:2$

Official Ans. by NTA (2)

**Sol.** 
$$dU = nC_v dT$$

$$dQ = nC_p dT$$

dW = PdV = nRdT (isobaric process)

$$dU:dQ:dW\ :\ C_v:C_p:R$$

$$=\frac{5R}{2}:\frac{7R}{2}:R=5:7:2$$

12. An engine of a train, moving with uniform acceleration, passes the signal-post with velocity u and the last compartment with velocity v. The velocity with which middle point of the train passes the signal post is:

(1) 
$$\sqrt{\frac{v^2 + u^2}{2}}$$

$$(2) \frac{v-u}{2}$$

$$(3) \ \frac{u+v}{2}$$

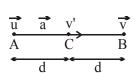
$$(4) \ \sqrt{\frac{v^2 - u^2}{2}}$$

Official Ans. by NTA (1)

**Sol.**  $(v')^2 = u^2 + 2ad$  $v^2 = (v')^2 + 2ad$ 

$$v^2 = (v')^2 + 2ad$$

solving, we get



$$v' = \sqrt{\frac{v^2 + u^2}{2}}$$

13. Match List-I with List-II:

List-I

List-II

- (a) h (Planck's constant)
- (i)  $[M L T^{-1}]$
- (b) E (kinetic energy)
- (ii) [M L<sup>2</sup> T<sup>-1</sup>]
- (c) V (electric potential)
- (iii) [M L<sup>2</sup> T<sup>-2</sup>]
- (d) P (linear momentum)
- (iv)[M  $L^2$  I<sup>-1</sup>T<sup>-3</sup>]

Choose the correct answer from the options given below:

- (1) (a) $\rightarrow$ (iii), (b) $\rightarrow$ (iv), (c) $\rightarrow$ (ii), (d) $\rightarrow$ (i)
- (2) (a) $\rightarrow$ (ii), (b) $\rightarrow$ (iii), (c) $\rightarrow$ (iv), (d) $\rightarrow$ (i)
- (3) (a) $\rightarrow$ (i), (b) $\rightarrow$ (ii), (c) $\rightarrow$ (iv), (d) $\rightarrow$ (iii)
- (4)  $(a)\rightarrow(iii)$ ,  $(b)\rightarrow(ii)$ ,  $(c)\rightarrow(iv)$ ,  $(d)\rightarrow(i)$

Official Ans. by NTA (2)

Sol. By dimensional analysis.

- **14.** Magnetic fields at two points on the axis of a circular coil at a distance of 0.05m and 0.2 m from the centre are in the ratio 8 : 1. The radius of coil is \_\_\_\_\_.
  - (1) 0.2 m
- (2) 0.1 m
- (3) 0.15 m
- (4) 1.0 m

## Official Ans. by NTA (2)

**Sol.** We know, the magnetic field on the axis of a current carrying circular ring is given by

$$B = \frac{\mu_0}{4\pi} \frac{2NIA}{\left(R^2 + x^2\right)^{3/2}}$$

$$\therefore \frac{B_1}{B_2} = \frac{8}{1} = \left[ \frac{R^2 + (0.2)^2}{R^2 + (0.05)^2} \right]^{3/2}$$

$$4[R^2 + (0.05)^2] = [R^2 + (0.2)^2]$$

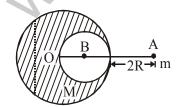
$$4R^2 - R^2 = (0.2)^2 - 4 \times (0.05)^2$$

$$4R^2 - R^2 = (0.2)^2 - (0.1)^2$$

$$3R^2 = 0.3 \times 0.1$$

$$R^2 = (0.1)^2 \Rightarrow R = 0.1$$

15. A solid sphere of radius R gravitationally attracts a particle placed at 3R form its centre with a force  $F_1$ . Now a spherical cavity of radius  $\left(\frac{R}{2}\right)$  is made in the sphere (as shown in figure) and the force becomes  $F_2$ . The value of  $F_1:F_2$  is :

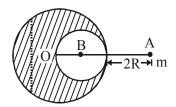


- (1) 25:36
- (2) 36:25
- (3) 50:41
- (4) 41 : 50

Official Ans. by NTA (3)

**Sol.** Let initial mass of sphere is m'. Hence mass of removed portion will be m'/8

$$F_1 = \text{m.E.} = \frac{\text{m.Gm'}}{9R^2}$$



$$F_2 = m \left[ \frac{G.m'}{(3R)^2} - \frac{G.m'/8}{(5R/2)^2} \right]$$

$$= \frac{Gm'}{9R^2} - \frac{Gm' \times 4}{8 \times 25}$$

$$= \left(\frac{1}{9} - \frac{1}{50}\right) \frac{Gm'}{R^2}$$

$$F_2 = \frac{41}{50 \times 9} \cdot \frac{Gm'}{R^2}$$

$$\frac{F_1}{F_2} = \frac{1}{9} \times \frac{50 \times 9}{41} = \frac{50}{41}$$

16. Two radioactive substances X and Y originally have N<sub>1</sub> and N<sub>2</sub> nuclei respectively. Half life of X is half of the half life of Y. After three half lives of Y, number of nuclei of both are equal.

The ratio  $\frac{N_1}{N_2}$  will be equal to :

 $(1) \frac{1}{8}$ 

(2)  $\frac{3}{1}$ 

(3)  $\frac{8}{1}$ 

 $(4) \frac{1}{3}$ 

Official Ans. by NTA (3)

**Sol.** 
$$T_x = t ; T_y = 2t$$

$$3T_{y} = 6t$$

$$N_1' = N_2'$$

$$N_1 e^{-\lambda_1 6t} = N_2 e^{-\lambda_2 6t}$$

$$\frac{N_1}{N_2} = e^{(\lambda_1 - \lambda_2)6t} = e^{\ln 2\left(\frac{1}{t} - \frac{1}{2t}\right) \times 6t} = e^{(\ln 2) \times 3} = e^{\ln 8} = 8$$

$$\frac{N_1}{N_2} = \frac{8}{1}$$

**17.** In an octagon ABCDEFGH of equal side, what is the sum of

$$\overrightarrow{AB} + \overrightarrow{AC} + \overrightarrow{AD} + \overrightarrow{AE} + \overrightarrow{AF} + \overrightarrow{AG} + \overrightarrow{AH}'$$

if, 
$$\overrightarrow{AO} = 2\hat{i} + 3\hat{j} - 4\hat{k}$$



- $(1) -16\hat{i} -24\hat{j} +32\hat{k}$
- (2)  $16\hat{i} + 24\hat{j} 32\hat{k}$
- (3)  $16\hat{i} + 24\hat{j} + 32\hat{k}$
- (4)  $16\hat{i} 24\hat{j} + 32\hat{k}$

Official Ans. by NTA (2)

Sol. We know,

 $\therefore \overrightarrow{OA} + \overrightarrow{OB} + \overrightarrow{OC} + \overrightarrow{OD} + \overrightarrow{OE} + \overrightarrow{OF} + \overrightarrow{OG} + \overrightarrow{OH} = \overrightarrow{0}$ By triangle law of vector addition, we can write

$$\overrightarrow{AB} = \overrightarrow{AO} + \overrightarrow{OB}$$
;  $\overrightarrow{AC} = \overrightarrow{AO} + \overrightarrow{OC}$ 

$$\overrightarrow{AD} = \overrightarrow{AO} + \overrightarrow{OD}$$
;  $\overrightarrow{AE} = \overrightarrow{AO} + \overrightarrow{OE}$ 

$$\overrightarrow{AF} = \overrightarrow{AO} + \overrightarrow{OF}$$
 ;  $\overrightarrow{AG} = \overrightarrow{AO} + \overrightarrow{OG}$ 

$$\overrightarrow{AH} = \overrightarrow{AO} + \overrightarrow{OH}$$

Now

$$\overrightarrow{AB} + \overrightarrow{AC} + \overrightarrow{AD} + \overrightarrow{AE} + \overrightarrow{AF} + \overrightarrow{AG} + \overrightarrow{AH}$$

$$= (7 \overline{\text{AO}}) + \overline{\text{OB}} + \overline{\text{OC}} + \overline{\text{OD}} + \overline{\text{OE}} + \overline{\text{OF}} + \overline{\text{OG}} + \overline{\text{OH}}$$

$$= (7 \overrightarrow{AO}) + \overrightarrow{O} - \overrightarrow{OA}$$

$$= (7 \overrightarrow{AO}) + \overrightarrow{AO}$$

$$= 8\overrightarrow{AO} = 8(2\hat{i} + 3\hat{j} - 4\hat{k})$$

$$= 16\hat{i} + 24\hat{j} - 32\hat{k}$$

**18.** Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason R.

Assertion A: The escape velocities of planet A and B are same. But A and

B are of unequal mass.

Reason R: The product of their mass and radius must be same,

 $M_1R_1 = M_2R_2$ 

In the light of the above statements, choose the most appropriate answer from the options given below:

- (1) Both A and R are correct but R is NOT the correct explanation of A
- (2) A is correct but R is not correct
- (3) Both A and R are correct and R is the correct explanation of A
- (4) A is not correct but R is correct

Official Ans. by NTA (2)

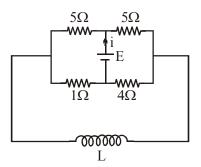
Sol. 
$$V_e = \sqrt{\frac{2GM}{R}}$$

$$\frac{\mathbf{M}_1}{\mathbf{R}_1} = \frac{\mathbf{M}_2}{\mathbf{R}_2}$$

$$M_1R_2 = M_2R_1$$

Hence reason R is not correct.

19. The current (i) at time t = 0 and  $t = \infty$  respectively for the given circuit is:



- (1)  $\frac{18E}{55}, \frac{5E}{18}$
- (2)  $\frac{10E}{33}, \frac{5E}{18}$
- (3)  $\frac{5E}{18}, \frac{18E}{55}$
- (4)  $\frac{5E}{18}, \frac{10E}{33}$

Official Ans. by NTA (4)

**Sol.** At t = 0, current through inductor is zero,

hence 
$$R_{eq} = (5+1) \parallel (5+4) = \frac{18}{5}$$

$$i_1 = \frac{E}{18/5} = \frac{5E}{18}$$

At  $t = \infty$ , inductor becomes a simple wire and now the circuit will be as shown in figure

hence 
$$R_{eq} = (5||5) + (4||1) = \frac{33}{10}$$
; ( $||\Rightarrow parallel$ )

$$i_2 = \frac{E}{33/10} = \frac{10E}{33}$$

20. Two coherent light sources having intensity in the ratio 2x produce an interference pattern.

The ratio  $\frac{I_{max} - I_{min}}{I_{mor} + I_{min}}$  will be:

(1) 
$$\frac{2\sqrt{2x}}{x+1}$$
 (2)  $\frac{\sqrt{2x}}{2x+1}$ 

$$(2) \ \frac{\sqrt{2x}}{2x+1}$$

$$(3) \ \frac{\sqrt{2x}}{x+1}$$

(4) 
$$\frac{2\sqrt{2x}}{2x+1}$$

Official Ans. by NTA (4)

**Sol.** Given that,  $\frac{I_1}{I_2} = 2x$ 

We know,

$$I_{\text{max}} = \left(\sqrt{I_1} + \sqrt{I_2}\right)^2 \& I_{\text{min}} = \left(\sqrt{I_1} - \sqrt{I_2}\right)^2$$

$$\therefore \frac{I_{\text{max}} - I_{\text{min}}}{I_{\text{max}} + I_{\text{min}}} = \frac{2\sqrt{I_1 I_2}}{I_1 + I_2} = \frac{2\sqrt{I_1 / I_2}}{\frac{I_1}{I_2} + 1} = \frac{2\sqrt{2x}}{2x + 1}$$

#### **SECTION-B**

1. A transmitting station releases waves of wavelength 960 m. A capacitor of 2.56 µF is used in the resonant circuit. The self inductance of coil necessary for resonance is  $_{---}$  × 10<sup>-8</sup> H.

Official Ans. by NTA (10)

**Sol.**  $\lambda = 960 \text{ m}$ 

$$C = 2.56 \ \mu F = 2.56 \times 10^{-6} \ F$$

$$c = 3 \times 10^8 \text{ m/s}$$

$$L = ?$$

Now at resonance,  $\omega_0 = \frac{1}{\sqrt{LC}}$ 

[Resoant frequency]

$$2\pi f_0 = \frac{1}{\sqrt{LC}}$$

On substituting  $f_0 = \frac{c}{\lambda}$ , we have  $2\pi \frac{c}{\lambda} = \frac{1}{\sqrt{I.C}}$ 

Squaring both sides :  $4\pi^2 \frac{c^2}{\lambda^2} = \frac{1}{1.00}$ 

$$=\frac{4\times10\times(3\times10^8)^2}{(960)^2}=\frac{1}{L\times2.56\times10^{-6}}$$

$$\Rightarrow \frac{1}{L} = \frac{4 \times 10 \times 9 \times 10^{16} \times 2.56 \times 10^{-6}}{960 \times 960}$$

$$\Rightarrow$$
 L = 10 × 10<sup>-8</sup> H

The electric field in a region is given

$$\vec{E} = \left(\frac{3}{5}E_0\hat{i} + \frac{4}{5}E_0\hat{j}\right)\frac{N}{C}$$
. The ratio of flux of

reported field through the rectangular surface of area  $0.2 \text{ m}^2$  (parallel to y - z plane) to that of the surface of area 0.3 m<sup>2</sup> (parallel to x - z plane) is a : b, where  $a = \underline{\hspace{1cm}}$ .

[Here  $\hat{i}, \hat{j}$  and  $\hat{k}$  are unit vectors along x, y and z-axes respectively]

Official Ans. by NTA (1)

**Sol.** 
$$\vec{E} = \left(\frac{3E_0}{5}\hat{i} + \frac{4E_0}{5}\hat{j}\right)\frac{N}{C}$$

$$A_1 = 0.2 \text{ m}^2$$
 [parallel to y - z plane]

$$= \vec{A}_1 = 0.2 \text{ m}^2 \hat{i}$$

$$A_2 = 0.3 \text{ m}^2$$
 [parallel to  $x - z$  plane]

$$\vec{A}_2 = 0.3 \, \text{m}^2 \, \hat{j}$$

Now 
$$\phi_a = \left[\frac{3E_0}{5}\hat{i} + \frac{4E_0}{5}\hat{j}\right] \cdot \left[0.2\,\hat{i}\right] = \frac{3\times0.2}{5}E_0$$

& 
$$\phi_b = \left[ \frac{3E_0}{5} \hat{i} + \frac{4E_0}{5} \hat{j} \right] \cdot \left[ 0.3 \, \hat{j} \right] = \frac{4 \times 0.3}{5} E_0$$

Now 
$$\frac{\phi_a}{\phi_b} = \frac{0.6}{1.2} = \frac{1}{2} = \frac{a}{b}$$

$$\Rightarrow$$
 a:b = 1:2

$$\Rightarrow$$
 a = 1

**3.** In a certain thermodynamical process, the pressure of a gas depends on its volume as kV<sup>3</sup>. The work done when the temperature changes from 100°C to 300°C will be \_\_\_\_ nR, where n denotes number of moles of a gas.

# Official Ans. by NTA (50)

**Sol.** 
$$P = kV^3$$

$$T_i = 100^{\circ}C$$
 &  $T_f = 300^{\circ}C$ 

$$\Delta T = 300 - 100$$

$$\Delta T = 200^{\circ} C$$

$$P = kV^3$$

now PV = nRT

$$\therefore kV^4 = nRT$$

$$now 4kV^3 dV = nRdT$$

$$\therefore$$
 PdV = nRdT/4

$$\therefore \text{ Work} = \int PdV = \int \frac{nRdT}{4} = \frac{nR}{4} \Delta T$$

$$= \frac{200}{4} \times nR = 50nR$$

4. A small bob tied at one end of a thin string of length 1m is describing a vertical circle so that the maximum and minimum tension in the string are in the ratio 5:1. The velocity of the bob at the height position is \_\_\_\_ m/s.

(Take 
$$g = 10 \text{ m/s}^2$$
)

## Official Ans. by NTA (5)

Let the speed of bob at lowest position be v<sub>1</sub> and at the highest position be  $v_2$ . Maximum tension is at lowest position and

minimum tension is at the highest position.

Now, using, conservation of mechanical energy,

$$\frac{1}{2}mv_1^2 = \frac{1}{2}mv_2^2 + mg \ 2l$$

$$\Rightarrow v_1^2 = v_2^2 + 4gl \dots (1)$$

Now 
$$T_{\text{max}}$$
-mg =  $\frac{\text{mv}_1^2}{I}$ 

$$\Rightarrow$$
 T<sub>max</sub> = mg +  $\frac{\text{mv}_1^2}{I}$ 

& 
$$T_{min} + mg = \frac{mv_2^2}{I}$$

$$\Rightarrow T_{\min} = \frac{mv_2^2}{I} - mg$$

$$\frac{T_{\text{max}}}{T_{\text{min}}} = \frac{5}{1}$$

$$\Rightarrow \frac{mg + \frac{mv_1^2}{l}}{\frac{mv_2^2}{l} - mg} = \frac{5}{1}$$

$$\Rightarrow mg + \frac{mv_1^2}{l} = \left\lceil \frac{mv_2^2}{l} - mg \right\rceil 5$$

$$\Rightarrow$$
 mg +  $\frac{m}{l} \left[ v_2^2 + 4g l \right] = \frac{5mv_2^2}{l} - 5mg$ 

$$\Rightarrow$$
 mg +  $\frac{\text{mv}_2^2}{l}$  + 4mg =  $\frac{5\text{mv}_2^2}{l}$  - 5mg

$$\Rightarrow 10 \text{mg} = \frac{4 \text{mv}_2^2}{l}$$

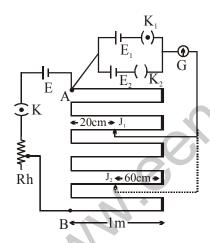
$$v_2^2 = \frac{10 \times 10 \times 1}{4}$$

$$\Rightarrow$$
  $v_2^2 = 25 \Rightarrow v_2 = 5 \text{ m/s}$ 

Thus, velocity of bob at highest position is 5 m/s.

5. In the given circuit of potentiometer, the potential difference E across AB (10m length) is larger than E<sub>1</sub> and E<sub>2</sub> as well. For key K<sub>1</sub> (closed), the jockey is adjusted to touch the wire at point J<sub>1</sub> so that there is no deflection in the galvanometer. Now the first battery (E<sub>1</sub>) is replaced by second battery (E<sub>2</sub>) for working by making K<sub>1</sub> open and K<sub>2</sub> closed. The galvanometer gives then null deflection at J<sub>2</sub>.

The value of  $\frac{E_1}{E_2}$  is  $\frac{a}{b}$ , where  $a = \underline{\hspace{1cm}}$ .



### Official Ans. by NTA (1)

**Sol.** Length of AB = 10 m

For battery  $E_1$ , balancing length is  $l_1$   $l_1 = 380$  cm [from end A]

For battery  $E_2$ , balancing length is  $l_2$   $l_2 = 760$  cm [from end A]

Now, we know that  $\frac{E_1}{E_2} = \frac{l_1}{l_2}$ 

$$\Rightarrow \frac{E_1}{E_2} = \frac{380}{760} = \frac{1}{2} = \frac{a}{b}$$

$$a = 1 & b = 2$$
.  
 $a = 1$ 

6. The same size images are formed by a convex lens when the object is placed at 20cm or at 10cm from the lens. The focal length of convex lens is \_\_\_\_ cm.

Official Ans. by NTA (15)

**Sol.** 
$$m = \frac{f}{u+f}$$

$$+ m = \frac{f}{-10+f}$$
 ... (1)

$$-m = \frac{f}{-20+f} \dots (2)$$

$$-1 = \frac{f - 20}{f - 10}$$

$$10 - f = f - 20$$

$$30 = 2f$$

$$f = 15 \text{ cm}$$

7. 512 identical drops of mercury are charged to a potential of 2V each. The drops are joined to form a single drop. The potential of this drop is \_\_\_\_ V.

Official Ans. by NTA (128)

**Sol.** 
$$Q = 512q$$

 $Volume_i = Volume_f$ 

$$512 \times \frac{4}{3} \pi r^3 = \frac{4}{3} \pi R^3$$

$$2^9 r^3 = R^3$$

$$R = 8r$$

$$2 = \frac{kq}{r}$$

$$V = \frac{kQ}{R} = \frac{k \ 5 \ 1 \ 2 \ q}{8 \ r}$$

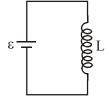
$$V = 128.$$

8. A coil of inductance 2H having negligible resistance is connected to a source of supply whose voltage is given by V = 3t volt. (where t is in second). If the voltage is applied when t = 0, then the energy stored in the coil after 4s is \_\_\_\_\_ J.

Official Ans. by NTA (144)

**Sol.** 
$$\varepsilon = \frac{L dI}{dt}$$

$$3\int_{0}^{4} t dt = 2\int_{0}^{I} dI$$



$$\frac{3}{2} \times 16 = 2I$$

$$I = 12$$

$$V = \frac{1}{2}LI^2 = \frac{1}{2} \times 2(12)^2 = 144 J$$

9. A monoatomic gas of mass 4.0 u is kept in an insulated container. Container is moving with velocity 30 m/s. If container is suddenly stopped then change in temperature of the gas

(R = gas constant) is 
$$\frac{x}{3R}$$
. Value of x is \_\_\_\_\_.

## Official Ans. by NTA (3600)

**Sol.** Given that mass of gas is 4u hence its molar mass M is 4g/mol

$$\therefore \frac{1}{2}mv^2 = nC_v \Delta T$$

$$\frac{1}{2}m \times (30)^2 = \frac{m}{M} \times \frac{3R}{2} \times \Delta T$$

$$\Delta T = \frac{3600}{3R}$$

**10.** The potential energy (U) of a diatomic molecule is a function dependent on r (interatomic distance) as

$$U = \frac{\alpha}{r^{10}} - \frac{\beta}{r^5} - 3$$

where,  $\alpha$  and  $\beta$  are positive constants. The equilibrium distance between two atoms will be

$$\left(\frac{2\alpha}{\beta}\right)^{\frac{a}{b}}$$
, where  $a =$ \_\_\_\_.

Official Ans. by NTA (1)

**Sol.** For equilibrium

$$\frac{dU}{dr} = 0$$

$$\frac{-10\alpha}{r^{11}} + \frac{5\beta}{r^6} = 0$$

$$\frac{5\beta}{r^6} = \frac{10\alpha}{r^{11}}$$

$$r^5 = \frac{2\alpha}{\beta}$$

$$r = \left(\frac{2\alpha}{\beta}\right)^{\frac{1}{5}}$$

a = 1

### **SECTION-A**

1. Given below are two statements:

Statement I : CeO<sub>2</sub> can be used for oxidation of aldehydes and ketones.

Statement II : Aqueous solution of EuSO<sub>4</sub> is a strong reducing agent.

In the light of the above statements, choose the correct answer from the options given below:

- (1) Statement I is false but statement II is true
- (2) Statment I is true but statement II is false
- (3) Both statement I and statement II are true
- (4) Both statement I and statement II are false **Official Ans. by NTA (3)**
- **Sol.** The +3 oxidation state of lanthanide is most stable and therefore lanthanide in +4 oxidation state has strong tendence to gain e<sup>-</sup> and converted into +3 and therefore act as strong oxidizing agent.

eg Ce<sup>+4</sup>

And there fore CeO<sub>2</sub> is used to oxidized alcohol aldehyde and ketones.

Lanthanide in +2 oxidation state has strong tendency to loss e and converted into +3 oxidation state therefore act as strong reducing agent.

- ∴ EuSO<sub>4</sub> act as strong reducing agent.
- 2. According to molecular theory, the species among the following that does not exist is: (1)  $He_2^+$  (2)  $He_2^-$  (3)  $Be_2$  (4)  $O_2^{2-}$

Official Ans. by NTA (3)

Sol.

Chemical Species	Bond Order
He <sub>2</sub> <sup>+</sup>	0.5
He <sub>2</sub>	0.5
Be <sub>2</sub>	0
$O_2^{2-}$	1

According to M.O.T. If bond order of chemical species is zero then that chemical species does not exist.

**3.** Which of the following reaction/s will not give p-aminoazobenzene?

(A) 
$$(i) \frac{\text{Sn/HCl}}{(ii) \text{HNO}_2}$$

$$(iii) \text{Aniline}$$

$$NO_2$$

(C) 
$$\frac{\text{(i) HNO}_2}{\text{(ii) Aniline, HCl}}$$

(1) A only

(2) B only

(3) C only

(4) A and B

### Official Ans. by NTA (2)

**Sol.** In basic or neutral medium N–N coupling favourable while in slightly acidic medium C–N coupling favourable.

(A) 
$$O_{2}$$
 (1)  $Sn + HCl$   $O_{2}$   $O$ 

(B) 
$$NO_2$$
 $NaBH_4$ 
 $No \ reaction$ 
 $NO_2$ 
 $NaOH$ 
 $No \ reaction$ 
 $No \ reaction$ 

(C) 
$$\stackrel{\text{NH}_2}{\longrightarrow} \stackrel{\text{NP}_2}{\longrightarrow} \stackrel{\text{NP}_2}{\longrightarrow} \stackrel{\text{NP}_2}{\longrightarrow} \stackrel{\text{NP}_2}{\longrightarrow} \stackrel{\text{NH}_2}{\longrightarrow} \stackrel{\text{NH}_2}$$

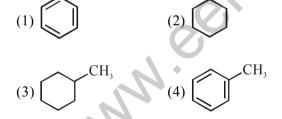
- 4. Which of the following equation depicts the oxidizing nature of H<sub>2</sub>O<sub>2</sub>?
  - (1)  $KIO_4 + H_2O_2 \rightarrow KIO_3 + H_2O + O_2$

  - (2)  $2I^{-} + H_{2}O_{2} + 2H^{+} \rightarrow I_{2} + 2H_{2}O$ (3)  $I_{2} + H_{2}O_{2} + 2OH^{-} \rightarrow 2I^{-} + 2H_{2}O + O_{2}$
  - (4)  $Cl_2 + H_2O_2 \rightarrow 2HCl + O_2$

Official Ans. by NTA (2)

- **Sol.** I- is oxidised to I<sub>2</sub> by H<sub>2</sub>O<sub>2</sub> Hence answer is (2)
- **5.** Identify A in the given chemical reaction.

$$\begin{array}{cccc} CH_3 & & & \\ CH_2 & CH_3 & & \underline{Mo_2O_3} & & \\ CH_2 & CH & & \overline{773~K,\,10\text{-}20~atm} & & \underline{major~product} \end{array}$$



Official Ans. by NTA (4)

**Sol.** 
$${}^{2}\sqrt{{}^{1}_{4}}$$
  ${}^{6}\sqrt{5}$   ${}^{10\text{-}20}$  atm  ${}^{+}$   ${}^{+}$   ${}^{+}$   ${}^{+}$   ${}^{+}$   ${}^{+}$ 

Mo<sub>2</sub>O<sub>3</sub> at 773 K temperature and 10-20-atm pressure is aromatising agent.

- 6. Complete combustion of 1.80 g of an oxygen containing compound (C<sub>x</sub>H<sub>y</sub>O<sub>z</sub>) gave 2.64 g of CO<sub>2</sub> and 1.08 g of H<sub>2</sub>O. The percentage of oxygen in the organic compound is:
  - (1) 51.63 (2) 63.53 (3) 53.33 (4) 50.33 Official Ans. by NTA (3)

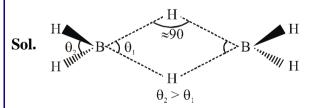
Sol. 
$$n_c = n_{co_2} = \frac{2.64}{44} = 0.06$$
  
 $n_H = 2 \times n_{H_2O} = \frac{1.08}{18} \times 2 = 0.12$   
 $m_0 = 1.80 - 12 \times \frac{2.64}{44} - \frac{1.08}{18} \times 2$   
 $= 1.80 - 0.72 - 0.12 = 0.96 \text{ gm}$   
 $\%0 = \frac{0.96}{1.80} \times 100 = 53.33\%$ 

Hence answer is (3)

- Which one of the following reactions will not 7. form acetaldehyde?
  - (1) CH<sub>3</sub>CH<sub>2</sub>OH -
  - (2) CH<sub>3</sub>CN (i) DIBAL (ii) H<sub>2</sub>O
  - (3)  $CH_2 = CH_2 + O_2 +$
  - (4)  $CH_3CH_2OH \xrightarrow{CrO_3-H_2SO_4}$

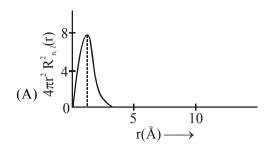
Official Ans. by NTA (4)

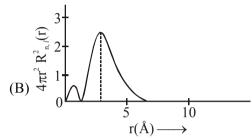
- 8. The correct statement about B<sub>2</sub>H<sub>6</sub> is:
  - (1) Terminal B-H bonds have less p-character when compared to bridging bonds.
  - (2) The two B–H–B bonds are not of same length
  - (3) All B-H-B angles are of 120°
  - (4) Its fragment, BH3, behaves as a Lewis base Official Ans. by NTA (1)

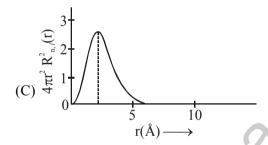


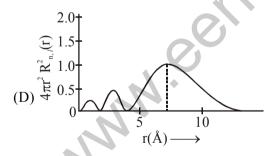
- $\theta_2 > \theta_1$ , : B-H (terminal) having less pcharacter as compare to bridge bond.
- Both B-H-B bridge bond having same bond length.
- B–H–B bond angle is ≈90°
- BH<sub>3</sub> is e<sup>-</sup> deficient species and therefore act as lewis acid

**9.** The plots of radial distribution functions for various orbitals of hydrogen atom against 'r' are given below:









The correct plot for 3s orbital is:

(1)(B)

(2)(A)

(3) (D)

(4) (C)

## Official Ans. by NTA (3)

**Sol.** Number of radial nodes =  $n - \ell - 1$ 

$$= 3 - 0 - 1 = 2$$

Therefor corresponding graph is (D)

Hence answer is (3)

**10.** Given below are two statements:

Statement I: An allotrope of oxygen is an important intermediate in the formation of reducing smog.

Statement II: Gases such as oxides of nitrogen and sulphur present in troposphere contribute to the formation of photochemical smog.

In the light of the above statements, choose the correct answer from the options given below:

(1) Both statement I and Statement II are false

(2) Statement I is true but Statement II is false

(3) Both Statement I and Statement II are true

(4) Statement I is false but Statement II is true

### Official Ans. by NTA (1)

**Sol.** Reducing smog is a mixture of smoke, fog and sulphur dioxide.

Tropospheric pollutants such as hydrocarbon and nitrogen oxide contirbute to the formation of photochemical smog.

11. In which of the following pairs, the outer most electronic configuration will be the same?

(1) Cr<sup>+</sup> and Mn<sup>2+</sup>

(2) Ni<sup>2+</sup> and Cu<sup>+</sup>

(3) Fe<sup>2+</sup> and Co<sup>+</sup>

(4) V2+ and Cr+

### Official Ans. by NTA (1)

**Sol.** Option -1  $Mn^{+2}[Ar]3d^5, Cr^+[Ar]3d^5$ Option -2  $Ni^{+2}[Ar]3d^8, Cu^+[Ar]3d^{10}$ Option -3  $Fe^{+2}[Ar]3d^6, Co^+[Ar]3d^74s^1$ Option -4  $V^{+2}[Ar]3d^3, Cr^+[Ar]3d^5$ 

**12.** Which of the glycosidic linkage between galactose and glucose is present in lactose?

(1) C-1 of galactose and C-4 of glucose

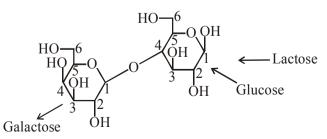
(2) C-1 of glucose and C-6 of galactose

(3) C-1 of glucose and C-4 of galactose

(4) C-1 of galactose and C-6 of glucose

#### Official Ans. by NTA (1)

Sol.



In lactose linkage is formed between  $C_1$  of galactose and  $C_4$  of gluocse.

**13.** Compound(s) which will liberate carbon dioxide with sodium bicarbonate solution is/are:

$$A = \underbrace{NH_2}_{OH} \underbrace{NH_2}_{NH_2} B = \underbrace{COOH}_{NH_2}$$

$$C = NO_2 \longrightarrow NO_2$$

$$NO_2$$

$$NO_2$$

- (1) B only
- (2) C only
- (3) B and C only
- (4) A and B only

## Official Ans. by NTA (3)

Sol. 
$$COOH$$
  $COONa$   $+ NaHCO_3 \Rightarrow O$   $+ H_2CO_3$  Weaker acid  $CO_2 \uparrow H_2CO_3 \uparrow CO_3 \uparrow$ 

equilibrium favours forward direction and  $CO_2 \uparrow$  is librated.

$$\begin{array}{c|c} OH & O \Theta_{Na} \Theta \\ NO_2 & O_2N & NO_2 \\ + NaHCO_3 & \longrightarrow & + H_2CO_3 \\ \hline NO_2 & & Weaker acid \\ NO_2 & & & \\ Stronger acid & & & \\ \hline \\ CO_2 \uparrow & H_2O \end{array}$$

Equilibrium favours forward direction and  $CO_2 \uparrow$  is librated.

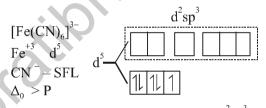
$$\begin{array}{c} NH_{2} \\ + NaHCO_{3} \Longrightarrow H_{2}CO_{3} + \\ NH_{2} \\ OH \\ Weak acid \\ \end{array}$$

Equilibrium favours back word direction and  $CO_2 \uparrow$  is not librated.

- 14. The hybridization and magnetic nature of  $[Mn(CN)_6]^{4-}$  and  $[Fe(CN)_6]^{3-}$ , respectively are:
  - (1) d<sup>2</sup>sp<sup>3</sup> and diamagnetic
  - (2) sp<sup>3</sup>d<sup>2</sup> and diamagnetic
  - (3) d<sup>2</sup>sp<sup>3</sup> and paramagnetic
  - (4) sp<sup>3</sup>d<sup>2</sup> and paramagnetic

Official Ans. by NTA (3)

- - .. hybridisation is d<sup>2</sup> sp<sup>3</sup> and due to presence of unpaired e<sup>-</sup> complex is paramagnetic in nature



- :. hybridisation is d<sup>2</sup>sp<sup>3</sup> and due to presence of unpaired e<sup>-</sup> complex paramagnetic in nature
- **15.** Ellingham diagram is a graphical representation of:
  - (1)  $\Delta H$  vs T
- (2)  $\Delta G$  vs T
- (3)  $\Delta G$  vs P
- (4)  $(\Delta G T\Delta S)$  vs T

Official Ans. by NTA (2)

- **Sol.** Ellingham diagram is a graphical representation of  $\Delta G$  vs T when metal heated with oxygen to form metal oxide
- **16.** The solubility of AgCN in a buffer solution of pH = 3 is x. The value of x is:

[Assume : No cyano complex is formed;  $K_{sp}(AgCN) = 2.2 \times 10^{-16}$  and  $K_{s}(HCN) = 6.2 \times 10^{-10}$ ]

- (1)  $0.625 \times 10^{-6}$
- $(2) 1.9 \times 10^{-5}$
- $(3) 2.2 \times 10^{-16}$
- (4)  $1.6 \times 10^{-6}$

Official Ans. by NTA (2)

Sol. 
$$\frac{K_{sp}}{Ka} = \frac{s^2}{(H^+)};$$
  $s = \sqrt{\frac{K_{sp}}{K_a}(H^+)}$  
$$s = \sqrt{\frac{2.2 \times 10^{-16}}{6.2 \times 10^{-10}} \times 10^{-3}}$$

$$s = 1.9 \times 10^{-5}$$

Hence answer is (2)

- 17. In Freundlich adsorption isotherm at moderate pressure, the extent of adsorption  $\left(\frac{x}{m}\right)$  is directly proportional to  $P^x$ . The value of x is
  - (1) zero
- (2)  $\frac{1}{n}$

(3) 1

(4) ∞

## Official Ans. by NTA (2)

Sol. As per Freundlich adsorption isotherm

$$\left(\frac{x}{m}\right) = KP^{\frac{1}{n}} \rightarrow x = \frac{1}{n}$$

Hence answer is (2)

**18.** Identify A and B in the chemical reaction.

$$\begin{array}{c}
OCH_3 \\
\hline
HCl \\
NO_2
\end{array}$$

$$\begin{array}{c}
IA] \\
(major) \\
\hline
MaI \\
dry acetone \\
(major)
\end{array}$$

$$\begin{array}{c}
IB] \\
(major)
\end{array}$$

$$(1) A = \bigvee_{NO_2}^{OCH_3} C$$

$$B = \bigcup_{NO_2}^{C1}$$

$$(2) A = \bigcup_{NO}^{OCH_3} CI$$

$$B = \bigvee_{NO_2} CI$$

(3) 
$$A = \bigcup_{M \in \mathcal{M}} CM$$

$$B = \bigvee_{NO_{7}}^{I} CI$$

(4) 
$$A = \bigvee_{NO_2}^{OCH_3} Cl$$

$$B = \bigcup_{NO_3}^{OCH_3} I$$

Official Ans. by NTA (4)

 $\Rightarrow$  I<sup>st</sup> reaction marcovnikov's addition of HCl on double bond while 2<sup>nd</sup> reaction is halide substitution by finkelstein reaction.

- **19.** Which statement is correct?
  - (1) Synthesis of Buna-S needs nascent oxygen.
  - (2) Neoprene is an addition copolymer used in plastic bucket manufacturing.
  - (3) Buna-S is a synthetic and linear thermosetting polymer.
  - (4) Buna-N is a natural polymer.

Official Ans. by NTA (1)

Sol. 
$$\uparrow$$
 Ph  $\xrightarrow{\text{Nascent}}$   $\uparrow$  CH<sub>2</sub>  $\xrightarrow{\text{CH}_2-\text{CH}_2-\text{CH}_2}$  BuNa-S Ph

**20.** The major product of the following chemical reaction is:

$$(1) H_3O^+, \Delta$$
  
 $CH_3CH_2CN \xrightarrow{(2) SOCl_2}$  ?

- (1) CH<sub>3</sub>CH<sub>2</sub>CH<sub>3</sub>
- (2) CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>OH
- (3) (CH<sub>3</sub>CH<sub>2</sub>CO)<sub>2</sub>O
- (4) CH<sub>3</sub>CH<sub>2</sub>CHO

Official Ans. by NTA (4)

Sol. Et-C=N
$$\xrightarrow{(i) H_3O^+/\Delta}$$
 Et-C-OH $\xrightarrow{(2) SOCl_2}$  Et-C-Cl
$$Et-C-H \xleftarrow{(3) Pd/BaSO_4}$$
Et-C-H  $\xleftarrow{H_2}$ 
reduction

Final product of reaction is propanaldehyde.

#### **SECTION-B**

- Among the following, the number of halide(s) 1. which is/are inert to hydrolysis is
  - (A) BF<sub>3</sub>
- (B) SiCl₄
- (C) PCl<sub>5</sub>
- (D) SF<sub>6</sub>

## Official Ans. by NTA (1)

- SF<sub>6</sub> is inert towards hydrolysis : answere is (1)
- 2. 1 molal aqueous solution of an electrolyte A<sub>2</sub>B<sub>3</sub> is 60% ionised. The boiling point of the solution at 1 atm is K. (Rounded-off to the nearest integer)

[Given  $K_b$  for  $(H_2O) = 0.52 \text{ K kg mol}^{-1}$ ]

Official Ans. by NTA (375)

**Sol.**  $\Delta T_b = iK_b m$  $= (1 + 4\alpha) \times 0.52 \times 1$  $= 3.4 \times 0.52 \times 1 = 1.768$  $T_{h} = 1.768 + 313.15 = 374.918 \text{ K}$ 

Hence answer is (375)

In basic medium  $CrO_4^{2-}$  oxidises  $S_2O_3^{2-}$  to form **3.**  $SO_4^{2-}$  and itself changes into  $Cr(OH)_4^-$ . The volume of 0.154 M CrO<sub>4</sub><sup>2-</sup> required to react with 40 mL of 0.25 M  $S_2O_3^{2-}$  is (Rounded-off to the nearest integer)

Official Ans. by NTA (173)

Sol. 
$$\overset{+6}{\text{CrO}_4^{2-}} + \overset{+2}{\text{S}_2} \overset{-2}{\text{O}_3^{2-}} \rightarrow \overset{+6}{\text{S}_4^{2-}} + \overset{+3}{\text{Cr}} (OH)_4^-$$
  
gm equi. of  $\text{CrO}_4^{2-} = S_2 \overset{-2}{\text{O}_3^{2-}}$   
 $0.154 \times 3 \times v = 0.25 \times 40 \times 8$   
 $v = 173.16 = 173 \text{ ml}$ 

Hence answer is (173)

4. A car tyre is filled with nitrogen gas at 35 psi at 27°C. It will burst if pressure exceeds 40 psi. The temperature in °C at which the car tyre will . (Rounded-off to the nearest burst is integer)

Official Ans. by NTA (70)

**Sol.**  $P \propto T$ 

$$\frac{P_2}{P_1} = \frac{T_2}{T_1} \Rightarrow \frac{40}{35} = \frac{T_2}{300}$$

$$T_2 = 342.854 \text{ K}$$
  
= 69.70°C \( \sim 70°C

Hence answer is (70)

**5**. The reaction of cyanamide, NH<sub>2</sub>CN<sub>(s)</sub> with oxygen was run in a bomb calorimeter and ΔU was found to be -742.24 kJ mol<sup>-1</sup>. The magnitude of  $\Delta H_{298}$  for the reaction

$${\rm NH_2CN_{(s)}} + \, \frac{3}{2} \, {\rm O_2(g)} \rightarrow {\rm N_{2(g)}} + \, {\rm O_2(g)} + \, {\rm H_2O_{(l)}}$$

is kJ. (Rounded off to the nearest integer)

[Assume ideal gases and  $R = 8.314 \text{ J mol}^{-1} \text{ K}^{-1}$ ]

Official Ans. by NTA (741)

**Sol.** 
$$\Delta H = \Delta U + \Delta n_{g}RT$$

$$= -742.24 + \frac{1}{2} \times \frac{8.314}{1000} \times 298$$

$$= -741 \text{ kJ/mol}$$

Hence answer is (741)

Using the provided information in the following paper chromatogram:

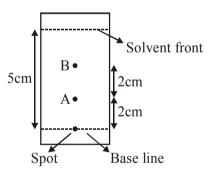


Figure: Paper chromatography for compounds A and B.

the calculate  $R_f$  value of A  $\times 10^{-1}$ .

Official Ans. by NTA (4)

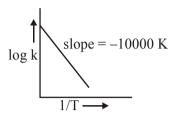
**Sol.**  $R_f = \frac{Distance travelled by compound}{Distance travelled by solvent}$ 

on chromatogram distance travelled by cmopound is  $\rightarrow$  2 cm

Distance travelled by solvent = 5 cm

So 
$$R_f = \frac{2}{5} = 4 \times 10^{-1} = 0.4$$

7. For the reaction,  $aA + bB \rightarrow cC + dD$ , the plot of log k vs  $\frac{1}{T}$  is given below:



The temperature at which the rate constant of the reaction is  $10^{-4}$  s<sup>-1</sup> is K.

(Rounded-off to the nearest integer)

[Given: The rate constant of the reaction is 10-5 s-1 at 500 K.]

Official Ans. by NTA (526)

**Sol.** 
$$\log K = \log A - \frac{Ea}{2.303RT}$$

$$|Slope| = \frac{Ea}{2.303R} = 10,000$$

$$\log\left(\frac{K_2}{K_1}\right) = \frac{Ea}{2.303R} \left(\frac{1}{T_1} - \frac{1}{T_2}\right)$$

$$\log\left(\frac{10^{-4}}{10^{-5}}\right) = 10,000 \left[\frac{1}{500} - \frac{1}{T_2}\right]$$

$$T_2 = 526.31 \approx 526K$$

Hence answer is (526)

8. 0.4 g mixture of NaOH, Na<sub>2</sub>CO<sub>3</sub> and some inert

impurities was first titrated with  $\frac{N}{10}$  HCl using

phenolphthalein as an indicator, 17.5 mL of HCl was required at the end point. After this methyl orange was added and titrated. 1.5 mL of same HCl was required for the next end point. The weight percentage of Na<sub>2</sub>CO<sub>3</sub> in the mixture is \_\_\_\_\_. (Rounded-off to the nearest integer)

Official Ans. by NTA (4)

Sol. Upto first end point gm equi. of  $(NaOH + Na_2CO_2) = HCl$ 

$$x + y \times 1 = \frac{1}{10} \times 17.5$$
  
 $x + y = 1.75$  ...(1)

Upto second end point

 $NaOH + Na_{2}CO_{3} \equiv HCl$ 

$$x + y \times 2 = \frac{1}{10} \times 19$$
  
 $x + 2y = 1.9$  ...(2)  
 $y = 0.15$ 

%Na<sub>2</sub>CO<sub>3</sub> = 
$$\frac{0.15 \times 10^{-3} \times 106}{0.4} \times 100$$
  
= 3.975%  
= 4%

$$= 3.975\%$$
  
= 4%

Hence answer is (4)

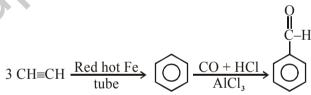
9. Consider the following chemical reaction.

CH 
$$\equiv$$
 CH  $\stackrel{\text{(1)}}{=}$  Red hot Fe tube, 873 K Product

The number of sp<sup>2</sup> hybridized carbon atom(s) present in the product is \_\_\_\_\_

Official Ans. by NTA (7)

Sol.



In benzaldehyde total number of sp<sup>2</sup> 'C' are 7.

10. The ionization enthalpy of Na<sup>+</sup> formation from Na<sub>(g)</sub> is 495.8 kJ mol<sup>-1</sup>, while the electron gain enthalpy of Br is -325.0 kJ mol<sup>-1</sup>. Given the lattice enthalpy of NaBr is -728.4 kJ mol<sup>-1</sup>. The energy for the formation of NaBr ionic solid is (-)  $\_\_\_ \times 10^{-1} \text{ kJ mol}^{-1}$ .

Official Ans. by NTA (5576)

Sol. 
$$Na(g) + Br(g) \longrightarrow NaBr(s)$$
 $IE_1 \downarrow \Delta Heg_1 \downarrow \qquad \qquad L.E.$ 
 $Na^+(g) + Br^-(g) \longrightarrow \Delta H_{formation} = IE_1 + \Delta Heg_1 + LE$ 
 $= 495.8 + (-325) + (-728.4)$ 
 $= -557.6$ 
 $= -5576 \times 10^{-1} \text{ KJ/mol.}$ 

Note: The above calculation is not for  $\Delta H_{formation}$  but for  $\Delta H_{Reaction}$ .

But on the basis of given data it is the best ans.

## **MATHEMATICS**

### **SECTION-A**

- 1. When a missile is fired from a ship, the probability that it is intercepted is  $\frac{1}{3}$  and the probability that the missile hits the target, given that it is not intercepted, is  $\frac{3}{4}$ . If three missiles are fired independently from the ship, then the probability that all three hit the target, is:
  - (1)  $\frac{1}{27}$  (2)  $\frac{3}{4}$  (3)  $\frac{1}{8}$  (4)  $\frac{3}{8}$

## Official Ans. by NTA (3)

- **Sol.** Required probability =  $\left(\frac{2}{3} \times \frac{3}{4}\right)^3 = \frac{1}{8}$
- If  $0 < \theta, \phi < \frac{\pi}{2}, x = \sum_{n=0}^{\infty} \cos^{2n} \theta, y = \sum_{n=0}^{\infty} \sin^{2n} \phi$ 2.

and 
$$z = \sum_{n=0}^{\infty} \cos^{2n} \theta \cdot \sin^{2n} \phi$$
 then :

- (1) xy z = (x + y) z (2) xy + yz + zx = z(3) xyz = 4 (4) xy + z = (x + y)z

### Official Ans. by NTA (4)

**Sol.**  $x = \frac{1}{1 - \cos^2 \theta} \implies \sin^2 \theta = \frac{1}{x}$ 

Also, 
$$\cos^2\theta = \frac{1}{y} \& 1 - \sin^2\theta \cos^2\theta = \frac{1}{z}$$

So, 
$$1 - \frac{1}{x} \times \frac{1}{y} = \frac{1}{z} \Rightarrow z(xy - 1) = xy$$
 ....(1)

Also, 
$$\frac{1}{x} + \frac{1}{y} = 1$$
  $\Rightarrow x + y = xy$  .....(2)

From (i) and (ii)

$$xy + z = xyz = (x + y) z$$

- **3.** Let f, g: N  $\rightarrow$  N such that f(n + 1) = f(n) + f(1) $\forall$  n  $\in$  N and g be any arbitrary function. Which of the following statements is NOT true?
  - (1) If fog is one-one, then g is one-one
  - (2) If f is onto, then  $f(n) = n \forall n \in \mathbb{N}$
  - (3) f is one-one
  - (4) If g is onto, then fog is one-one

#### Official Ans. by NTA (4)

- **Sol.** f(n + 1) f(n) = f(1)
  - $\Rightarrow$  f(n) = nf (1)

 $\Rightarrow$  f is one-one

Now, Let  $f(g(x_2)) = f(g(x_1))$ 

$$\Rightarrow$$
 g(x<sub>2</sub>) = g(x<sub>1</sub>) (as f is one-one)

$$\Rightarrow$$
  $x_1 = x_2$  (as fog is one-one)

 $\Rightarrow$  g is one-one

Now, f(g(n)) = g(n) f(1)

may be many-one if

g(n) is many-one

The equation of the line through the point (0,1,2) and perpendicular to the line

$$\frac{x-1}{2} = \frac{y+1}{3} = \frac{z-1}{-2}$$
 is:

(1) 
$$\frac{x}{3} = \frac{y-1}{4} = \frac{z-2}{3}$$

(2) 
$$\frac{x}{3} = \frac{y-1}{-4} = \frac{z-2}{3}$$

(3) 
$$\frac{x}{3} = \frac{y-1}{4} = \frac{z-2}{-3}$$

(4) 
$$\frac{x}{-3} = \frac{y-1}{4} = \frac{z-2}{3}$$

## Official Ans. by NTA (4)

**Sol.**  $\frac{x-1}{2} = \frac{y+1}{3} = \frac{z-1}{-2} = r$ 

$$\Rightarrow$$
 P(x, y, z) = (2r + 1, 3r - 1, -2r + 1)

Since,  $\overrightarrow{QP} \perp (2\hat{i} + 3\hat{j} - 2\hat{k})$ 

$$\Rightarrow$$
 4r + 2 + 9r - 6 + 4r + 2 = 0

$$\Rightarrow r = \frac{2}{17}$$

$$\Rightarrow P\left(\frac{21}{17}, \frac{-11}{17}, \frac{13}{17}\right)$$



$$\Rightarrow \overrightarrow{PQ} = \frac{21\hat{i} - 28\hat{j} - 21\hat{k}}{17}$$

So, 
$$\overrightarrow{QP}: \frac{x}{-3} = \frac{y-1}{4} = \frac{z-2}{3}$$

5. Let  $\alpha$  be the angle between the lines whose direction cosines satisfy the equations l + m - n = 0 and  $l^2 + m^2 - n^2 = 0$ . Then the value of  $\sin^4\alpha + \cos^4\alpha$  is :

(1) 
$$\frac{3}{4}$$

(1) 
$$\frac{3}{4}$$
 (2)  $\frac{3}{8}$  (3)  $\frac{5}{8}$ 

(3) 
$$\frac{5}{8}$$

$$(4) \frac{1}{2}$$

### Official Ans. by NTA (3)

**Sol.**  $n = \ell + m$ 

Now, 
$$\ell^2 + m^2 = n^2 = (\ell + m)^2$$
  
 $\Rightarrow 2\ell m = 0$ 

If 
$$\ell = 0 \implies m = n = \pm \frac{1}{\sqrt{2}}$$

And, If 
$$m = 0 \Rightarrow n = \ell = \pm \frac{1}{\sqrt{2}}$$

So, direction cosines of two lines are

$$\left(0, \frac{1}{\sqrt{2}}, \frac{1}{\sqrt{2}}\right)$$
 and  $\left(\frac{1}{\sqrt{2}}, 0, \frac{1}{\sqrt{2}}\right)$ 

Thus, 
$$\cos \alpha = \frac{1}{2} \Rightarrow \alpha = \frac{\pi}{3}$$

6. The value of the integral

$$\int \frac{\sin\theta.\sin2\theta(\sin^6\theta+\sin^4\theta+\sin^2\theta)\sqrt{2\sin^4\theta+3\sin^2\theta+6}}{1-\cos2\theta}d\theta$$

is:

(where c is a constant of integration)

$$(1) \ \frac{1}{18} \Big[ 11 - 18\sin^2\theta + 9\sin^4\theta - 2\sin^6\theta \Big]^{\frac{3}{2}} + c$$

(2) 
$$\frac{1}{18} \left[ 9 - 2\cos^6\theta - 3\cos^4\theta - 6\cos^2\theta \right]^{\frac{3}{2}} + c$$

(3) 
$$\frac{1}{18} \left[ 9 - 2\sin^6\theta - 3\sin^4\theta - 6\sin^2\theta \right]^{\frac{3}{2}} + c$$

(4) 
$$\frac{1}{18} \left[ 11 - 18\cos^2\theta + 9\cos^4\theta - 2\cos^6\theta \right]^{\frac{3}{2}} + c$$

Official Ans. by NTA (4)

$$\textbf{Sol.} \quad I = \int \frac{\sin\theta.\sin2\theta \left(\sin^6\theta + \sin^4\theta + \sin^2\theta\right) \sqrt{2\sin^4\theta + 3\sin^2\theta + 6}}{1 - \cos2\theta} \ d\theta$$

$$\Longrightarrow I = \int \frac{\sin\theta \cdot 2\sin\theta \cos\theta \cdot \sin^2\theta \left(\sin^4\theta + \sin^2\theta + 1\right) \left(2\sin^4\theta + 3\sin^2\theta + 6\right)^{1/2}}{2\sin^2\theta} \, d\theta$$

$$= \int \sin^2\theta \cdot \cos\theta \big(\sin^4\theta + \sin^2\theta + 1\big) \big(2\sin^4\theta + 3\sin^2\theta + 6\big)^{1/2} \; d\theta$$

Let  $\sin\theta = t \Rightarrow \cos\theta \ d\theta = dt$ 

$$I = \int t^2 (t^4 + t^2 + 1) (2t^4 + 3t^2 + 6)^{1/2} dt$$
$$= \int (t^5 + t^3 + t) t (2t^4 + 3t^2 + 6)^{1/2} dt$$

$$= \int (t^5 + t^3 + t)(t^2)^{1/2} (2t^4 + 3t^2 + 6)^{1/2} dt$$
$$= \int (t^5 + t^3 + t)(2t^6 + 3t^4 + 6t^2)^{1/2} dt$$

Let 
$$2t^6 + 3t^4 + 6t^2 = u^2$$
  
 $\Rightarrow 12(t^5 + t^3 + t) dt = 2udu$ 

$$I = \int (u^2)^{1/2} \cdot \frac{2udu}{12}$$

$$= \int \frac{u^2}{6} du = \frac{u^3}{18} + C$$

$$= \frac{(2t^6 + 3t^4 + 6t^2)^{3/2}}{18} + C$$

when  $t = \sin\theta$ and  $t^2 = 1 - \cos^2\theta$  will give option (4)

The value of  $\int x^2 e^{\left[x^3\right]} dx$ , where [t] denotes the greatest integer  $\leq$  t, is:

(1) 
$$\frac{e-1}{3e}$$
 (2)  $\frac{e+1}{3}$  (3)  $\frac{e+1}{3e}$  (4)  $\frac{1}{3e}$ 

Official Ans. by NTA (3)

**Sol.** 
$$I = \int_{-1}^{1} x^2 e^{[x^3]} dx$$

$$= \int\limits_{-1}^{0} x^2 e^{[x^3]} dx + \int\limits_{0}^{1} x^2 e^{[x^3]} dx$$

$$= \int_{1}^{0} x^{2} e^{-1} dx + \int_{0}^{1} x^{2} e^{0} dx$$

$$=\frac{1}{e} \times \frac{x^3}{3} \begin{vmatrix} 0 \\ -1 + \frac{x^3}{3} \end{vmatrix}_{0}^{1}$$

$$=\frac{1}{e}\times\left(0-\left(\frac{-1}{3}\right)\right)+\frac{1}{3}$$

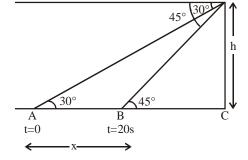
$$=\frac{1}{3e}+\frac{1}{3}=\frac{1+e}{3e}$$

8. A man is observing, from the top of a tower, a boat speeding towards the tower from a certain point A, with uniform speed. At that point, angle of depression of the boat with the man's eye is  $30^{\circ}$  (Ignore man's height). After sailing for 20 seconds, towards the base of the tower (which is at the level of water), the boat has reached a point B, where the angle of depression is  $45^{\circ}$ . Then the time taken (in seconds) by the boat from B to reach the base of the tower is:

- (1) 10
- (2)  $10\sqrt{3}$
- (3)  $10(\sqrt{3}+1)$
- (4)  $10(\sqrt{3}-1)$

Official Ans. by NTA (3)

Sol.



Let speed of boat is u m/s and height of tower is h meter & distance AB = x metre

$$\therefore$$
 x = h cot 30° - h cot 45°

$$\Rightarrow$$
 x = h  $(\sqrt{3}-1)$ 

$$\therefore u = \frac{x}{20} = \frac{h(\sqrt{3} - 1)}{20} \text{ m/s}$$

∴ Time taken to travel from B to C (Distance = h meter)

$$= \frac{h}{u} = \frac{h}{h \frac{(\sqrt{3} - 1)}{20}} = \frac{20}{\sqrt{3} - 1} = 10(\sqrt{3} + 1)\sec x$$

- 9. A tangent is drawn to the parabola  $y^2 = 6x$  which is perpendicular to the line 2x + y = 1. Which of the following points does NOT lie on it?
  - (1) (-6, 0) (2) (4, 5) (3) (5, 4) (4) (0, 3) **Official Ans. by NTA (3)**
- **Sol.** Slope of tangent =  $m_T = m$

So, m (-2) = -1 
$$\Rightarrow$$
 m =  $\frac{1}{2}$ 

Equation : 
$$y = mx + \frac{a}{m}$$

$$\Rightarrow y = \frac{1}{2}x + \frac{3}{2 \times \frac{1}{2}} \left( a = \frac{6}{4} = \frac{3}{2} \right)$$

$$\Rightarrow y = \frac{x}{2} + 3$$

$$\Rightarrow$$
 2y = x + 6

Point (5, 4) will not lie on it

10. All possible values of  $\theta \in [0, 2\pi]$  for which  $\sin 2\theta + \tan 2\theta > 0$  lie in :

$$(1) \left(0, \frac{\pi}{2}\right) \cup \left(\pi, \frac{3\pi}{2}\right)$$

$$(2) \left(0, \frac{\pi}{2}\right) \cup \left(\frac{\pi}{2}, \frac{3\pi}{4}\right) \cup \left(\pi, \frac{7\pi}{6}\right)$$

$$(3) \left(0, \frac{\pi}{4}\right) \cup \left(\frac{\pi}{2}, \frac{3\pi}{4}\right) \cup \left(\frac{3\pi}{2}, \frac{11\pi}{6}\right)$$

$$(4) \left(0,\frac{\pi}{4}\right) \cup \left(\frac{\pi}{2},\frac{3\pi}{4}\right) \cup \left(\pi,\frac{5\pi}{4}\right) \cup \left(\frac{3\pi}{2},\frac{7\pi}{4}\right)$$

Official Ans. by NTA (4)

**Sol.** 
$$\sin 2\theta + \tan 2\theta > 0$$

$$\Rightarrow \sin 2\theta + \frac{\sin 2\theta}{\cos 2\theta} > 0$$

$$\Rightarrow \sin 2\theta \frac{(\cos 2\theta + 1)}{\cos 2\theta} > 0 \Rightarrow \tan 2\theta (2\cos^2 \theta) > 0$$

Note:  $\cos 2\theta \neq 0$ 

$$\Rightarrow 1-2 \sin^2\theta \neq 0 \Rightarrow \sin\theta \neq \pm \frac{1}{\sqrt{2}}$$

Now,  $\tan 2\theta (1 + \cos 2\theta) > 0$ 

$$\Rightarrow \tan 2\theta > 0$$

(as 
$$\cos 2\theta + 1 > 0$$
)

$$\Rightarrow 2\theta \in \left(0, \frac{\pi}{2}\right) \cup \left(\pi, \frac{3\pi}{2}\right) \cup \left(2\pi, \frac{5\pi}{2}\right) \cup \left(3\pi, \frac{7\pi}{2}\right)$$

$$\Rightarrow \theta \in \left(0, \frac{\pi}{4}\right) \cup \left(\frac{\pi}{2}, \frac{3\pi}{4}\right) \cup \left(\pi, \frac{5\pi}{4}\right) \cup \left(\frac{3\pi}{2}, \frac{7\pi}{4}\right)$$

As  $\sin \theta \neq \pm \frac{1}{\sqrt{2}}$ ; which has been already

considered

- Let the lines  $(2 i)z = (2 + i)\overline{z}$  and 11.  $(2 + i)z + (i - 2)\overline{z} - 4i = 0$ , (here  $i^2 = -1$ ) be normal to a circle C. If the line  $iz + \overline{z} + 1 + i = 0$ is tangent to this circle C, then its radius is:

- (1)  $\frac{3}{\sqrt{2}}$  (2)  $\frac{1}{2\sqrt{2}}$  (3)  $3\sqrt{2}$  (4)  $\frac{3}{2\sqrt{2}}$

Official Ans. by NTA (4)

**Sol.** (i)  $(2 - i)z = (2 + i) \overline{z}$ 

$$y = \frac{x}{2}$$

(ii)  $(2 + i) z + (i - 2) \overline{z} - 4i = 0$ 

$$x + 2y = 2$$

(iii)  $iz + \overline{z} + 1 + i = 0$ 

Eqn of tangent x-y+1=0

Solving (i) and (ii)

$$x = 1, y = \frac{1}{2}$$

Now, 
$$p = r \Rightarrow \left| \frac{1 - \frac{1}{2} + 1}{\sqrt{2}} \right| = r$$

$$\Rightarrow r = \frac{3}{2\sqrt{2}}$$

The image of the point (3, 5) in the line **12.** x - y + 1 = 0, lies on :

$$(1) (x - 2)^2 + (y - 2)^2 = 12$$

(2) 
$$(x - 4)^2 + (y + 2)^2 = 16$$

(3) 
$$(x - 4)^2 + (y - 4)^2 = 8$$

$$(4) (x - 2)^2 + (y - 4)^2 = 4$$

Official Ans. by NTA (4)

Sol. 
$$P(3,5)$$
 $x-y+1=0$ 
 $P^{1}(x,y)$ 

$$\frac{x-3}{1} = \frac{y-5}{-1} = -2\left(\frac{3-5+1}{1+1}\right)$$

So, x = 4, y = 4Hence,  $(x - 2)^2 + (y - 4)^2 = 4$ 

If the curves,  $\frac{x^2}{a} + \frac{y^2}{b} = 1$  and  $\frac{x^2}{c} + \frac{y^2}{d} = 1$ 

intersect each other at an angle of 90°, then which of the following relations is TRUE?

$$(1) a + b = c + d$$

$$(2) a - b = c - d$$

(3) 
$$a - c = b + d$$

$$(4) ab = \frac{c+d}{a+b}$$

Official Ans. by NTA (2)

**Sol.** For orthogonal curves a - c = b - d $\Rightarrow$  a - b = c - d

14. 
$$\lim_{n\to\infty} \left(1 + \frac{1+\frac{1}{2} + \dots + \frac{1}{n}}{n^2}\right)^n$$
 is equal to:

- (1)  $\frac{1}{2}$  (2) 0 (3)  $\frac{1}{2}$
- (4) 1

Official Ans. by NTA (4)

**Sol.** Given limit is of  $1^{\infty}$  form

So, 
$$l = \exp\left(\lim_{n \to \infty} \frac{1 + \frac{1}{2} + \frac{1}{3} + \dots + \frac{1}{n}}{n}\right)$$

Now,

$$0 \le 1 + \frac{1}{2} + \frac{1}{3} + \dots + \frac{1}{n} \le 1 + \frac{1}{\sqrt{2}} + \frac{1}{\sqrt{3}} + \dots + \frac{1}{\sqrt{n}}$$

$$\leq 2\sqrt{n} - 1$$

So,  $l = \exp(0)$  (from sandwich theorem) = 1

- **15.** The coefficients a, b and c of the quadratic equation,  $ax^2 + bx + c = 0$  are obtained by throwing a dice three times. The probability that this equation has equal roots is:

  - (1)  $\frac{1}{72}$  (2)  $\frac{5}{216}$  (3)  $\frac{1}{36}$  (4)  $\frac{1}{54}$

## Official Ans. by NTA (2)

Sol.  $ax^2 + bx + c = 0$ 

For equal roots D = 0

 $\Rightarrow$  b<sup>2</sup> = 4ac

Case I: ac = 1

(a, b, c) = (1, 2, 1)

Case II: ac = 4

(a, b, c) = (1, 4, 4)

or (4, 4, 1)

or (2, 4, 2)

Case III: ac = 9

(a, b, c) = (3, 6, 3)

Required probability =  $\frac{5}{216}$ 

**16.** The total number of positive integral solutions (x, y, z) such that xyz = 24 is:

(3) 45

- (1) 36
- (2) 24
- (4) 30

#### Official Ans. by NTA (4)

**Sol.**  $xyz = 2^3 \times 3^1$ 

Let  $x = 2^{\alpha_1} \times 3^{\beta_1}$ 

 $v = 2^{\alpha_2} \times 3^{\beta_2}$ 

 $z = 2^{\alpha_3} \times 3^{\beta_2}$ 

Now  $\alpha_1 + \alpha_2 + \alpha_3 = 3$ .

No. of non-negative intergal sol =  ${}^{5}C_{2} = 10$ 

&  $\beta_1 + \beta_2 + \beta_3 = 1$ 

No. of non-negative intergal sol<sup>n</sup> =  ${}^{3}C_{2} = 3$ 

Total ways =  $10 \times 3 = 30$ .

- The integer 'k', for which the inequality **17.**  $x^2 - 2(3k - 1)x + 8k^2 - 7 > 0$  is valid for every x in R, is:
  - (1) 3
- (2) 2
- (3) 0
- (4) 4

#### Official Ans. by NTA (1)

**Sol.**  $x^2 - 2(3K - 1) x + 8K^2 - 7 > 0$ 

Now. D < 0

 $\Rightarrow$  4 (3K - 1)<sup>2</sup> - 4 × 1 × (8K<sup>2</sup> - 7) < 0

 $\Rightarrow$  9 K<sup>2</sup> - 6 K + 1 - 8K<sup>2</sup> + 7 < 0

 $\Rightarrow$  K<sup>2</sup> – 6K + 8 < 0

 $\Rightarrow$  (K – 4) (K – 2) < 0

 $\Rightarrow |K \in (2,4)|$ 

**18.** If a curve passes through the origin and the slope of the tangent to it at any point (x, y) is

 $\frac{x^2-4x+y+8}{x-2}$ , then this curve also passes

through the point:

- (1)(5,4)
- (2)(4,5)
- (3)(4,4)
- (4) (5, 5)

Official Ans. by NTA (4)

Sol. Given

$$y(0) = 0$$

& 
$$\frac{dy}{dx} = \frac{(x-2)^2 + y + 4}{x-2}$$

$$\Rightarrow \frac{dy}{dx} - \frac{y}{x-2} = (x-2) + \frac{4}{x-2}$$

$$\Rightarrow$$
 I.F. =  $e^{-\int \frac{1}{x-2} dx} = \frac{1}{x-2}$ 

Solution of L.D.E.

$$\Rightarrow y \frac{1}{x-2} = \int \frac{1}{x-2} \left( (x-2) + \frac{4}{x-2} \right) . dx$$

$$\Rightarrow \frac{y}{x-2} = x - \frac{4}{x-2} + C$$

Now, at x = 0,  $y = 0 \Rightarrow C = -2$ 

$$y = x (x - 2) - 4 - 2 (x - 2)$$

$$\Rightarrow$$
 y =  $x^2 - 4x$ 

This curve passes through (5, 5)

- **19.** The statement  $A \rightarrow (B \rightarrow A)$  is equivalent to :
  - $(1) A \rightarrow (A \land B)$
- $(2) A \rightarrow (A \rightarrow B)$
- $(3) A \rightarrow (A \leftrightarrow B)$
- $(4) A \rightarrow (A \lor B)$

Official Ans. by NTA (4)

- **Sol.**  $A \rightarrow (B \rightarrow A)$ 
  - $\equiv A \rightarrow (\sim B \lor A)$
  - $\equiv \sim A \lor (\sim B \lor A)$
  - $\equiv (\sim A \vee A) \vee \sim B$
  - $\equiv T \lor \sim B \equiv T$
  - $T \lor B = T$
  - $\equiv (\sim A \lor A) \lor B$
  - $\equiv \sim A \lor (A \lor B)$
  - $\equiv A \rightarrow (A \lor B)$

- **20.** If Rolle's theorem holds for the function  $f(x) = x^3 ax^2 + bx 4$ ,  $x \in [1, 2]$  with  $f'\left(\frac{4}{3}\right) = 0$ , then ordered pair (a, b) is equal to:
  - (1) (5, 8)
- (2) (-5, 8)
- (3) (5, -8)
- (4) (-5, -8)

## Official Ans. by NTA (1)

- **Sol.** f(1) = f(2)  $\Rightarrow 1 - a + b - 4 = 8 - 4a + 2b - 4$   $\Rightarrow 3a - b = 7$  .....(1)
  - Also  $f^1\left(\frac{4}{3}\right) = 0$  (given)
  - $\Rightarrow \left(3x^2 2ax + b\right)_{x = \frac{4}{3}} = 0$
  - $\Rightarrow \frac{16}{3} \frac{8a}{3} + b = 0$
  - $\Rightarrow 8a 3b 16 = 0$  ....(2)

Solving (1) and (2)

a = 5, b = 8

#### **SECTION-B**

1. Let f(x) be a polynomial of degree 6 in x, in which the coefficient of  $x^6$  is unity and it has

extrema at x = -1 and x = 1. If  $\lim_{x \to 0} \frac{f(x)}{x^3} = 1$ , then

 $5 \cdot f(2)$  is equal to \_\_\_\_\_.

### Official Ans. by NTA (144)

**Sol.** Let  $f(x) = x^6 + ax^5 + bx^4 + cx^3 + dx^2 + ex + f$ 

as  $\lim_{x\to 0} \frac{f(x)}{x^3} = 1$  non-zero finite

So, d = e = f = 0

and  $f(x) = x^3(x^3 + ax^2 + bx + c)$ 

Hence,  $\lim_{x\to 0} \frac{f(x)}{x^3} = c = 1$ 

Now, as  $f(x) = x^6 + ax^5 + bx^4 + x^3$ 

and f'(x) = 0 at x = 1 and x = -1

i.e.,  $f'(x) = 6x^5 + 5ax^4 + 4bx^3 + 3x^2$ 

f'(1) = 0

 $\Rightarrow 6 + 5a + 4b + 3 = 0$ 

 $\Rightarrow$  5a + 4b = -9

- & f'(-1) = 0
- $\Rightarrow -6 + 5a 4b + 3 = 0$
- $\Rightarrow$  5a 4b = 3

Solving both we get,

- $a = \frac{-6}{10} = \frac{-3}{5}$ ;  $b = \frac{-3}{2}$
- $\therefore f(x) = x^6 \frac{3}{5}x^5 \frac{3}{2}x^4 + x^3$
- $\therefore 5f(2) = 5 \left[ 64 \frac{3}{5} \cdot 32 \frac{3}{2} \cdot 16 + 8 \right]$  = 320 96 120 + 40 = 144
- 2. The number of points, at which the function f(x)=  $|2x + 1| - 3|x + 2| + |x^2 + x - 2|$ ,  $x \in R$  is not differentiable, is \_\_\_\_\_.

Official Ans. by NTA (2)

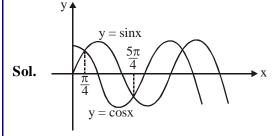
Sol.  $f(x) = |2x+1|-3|x+2|+|x^2+x-2|$ = |2x+1|-3|x+2|+|x+2||x-1|= |2x+1|+|x+2|(|x-1|-3)

Critical points are  $x = \frac{-1}{2}, -2, -1$ 

but x = -2 is making a zero. twice in product so, points of non differentability are  $x = \frac{-1}{2}$  and x = -1

- $\therefore$  Number of points of non-differentiability =  $\boxed{2}$
- 3. The graphs of sine and cosine functions, intersect each other at a number of points and between two consecutive points of intersection, the two graphs enclose the same area A. Then A<sup>4</sup> is equal to \_\_\_\_\_.

Official Ans. by NTA (64)



$$A = \int_{\pi/4}^{5\pi/4} (\sin x - \cos x) dx$$

$$= (-\cos x - \sin x)\Big|_{\pi/4}^{5\pi/4}$$

$$= \left(-\left(\frac{-1}{\sqrt{2}}\right) - \left(\frac{-1}{\sqrt{2}}\right)\right) - \left(-\left(\frac{1}{\sqrt{2}}\right) - \left(\frac{1}{\sqrt{2}}\right)\right)$$

$$\Rightarrow A = \frac{2}{\sqrt{2}} + \frac{2}{\sqrt{2}} = 2\sqrt{2}$$

$$\Rightarrow A^4 = (2\sqrt{2})^4 = 16 \times 4 = 64$$

**4.** Let  $A_1$ ,  $A_2$ ,  $A_3$ , ...... be squares such that for each  $n \ge 1$ , the length of the side of  $A_n$  equals the length of diagonal of  $A_{n+1}$ . If the length of  $A_1$  is 12 cm, then the smallest value of n for which area of  $A_n$  is less than one, is \_\_\_\_\_.

Official Ans. by NTA (9)

**Sol.** Let  $a_n$  be the side length of  $A_n$ .

So, 
$$a_n = \sqrt{2}a_{n+1}$$
,  $a_1 = 12$ 

$$\Rightarrow a_n = 12 \times \left(\frac{1}{\sqrt{2}}\right)^{n-1}$$

Now, 
$$(a_n)^2 < 1 \implies \frac{144}{2^{(n-1)}} < 1$$

$$\Rightarrow 2^{(n-1)} > 144$$

$$\Rightarrow$$
 n - 1  $\geq$  8

$$\Rightarrow$$
 n  $\geq$  9

5. Let  $A = \begin{bmatrix} x & y & z \\ y & z & x \\ z & x & y \end{bmatrix}$ , where x, y and z are real

numbers such that x + y + z > 0 and xyz = 2. If  $A^2 = I_3$ , then the value of  $x^3 + y^3 + z^3$  is\_\_\_\_\_.

Official Ans. by NTA (7)

**Sol.** 
$$A^2 = I$$

$$\Rightarrow AA' = I (as A' = A)$$

 $\Rightarrow$  A is orthogonal

So, 
$$x^2 + y^2 + z^2 = 1$$
 and  $xy + yz + zx = 0$ 

$$\Rightarrow$$
 (x + y + z)<sup>2</sup> = 1 + 2 × 0

$$\Rightarrow$$
 x + y + z = 1

Thus,

$$x^3 + y^3 + z^3 = 3 \times 2 + 1 \times (1 - 0)$$
  
= 7

**6.** If 
$$A = \begin{bmatrix} 0 \\ \tan(\frac{\theta}{2}) \\ 0 \end{bmatrix}$$
 and

$$(I_2 + A) (I_2 - A)^{-1} = \begin{bmatrix} a & -b \\ b & a \end{bmatrix}$$
, then 13  $(a^2 + b^2)$ 

is equal to

Official Ans. by NTA (13)

Sol. 
$$a^2 + b^2 = |I_2 + A||I_2 - A|^{-1}$$
  
=  $\sec^2 \frac{\theta}{2} \times \cos^2 \frac{\theta}{2} = 1$ 

7. The total number of numbers, lying between 100 and 1000 that can be formed with the digits 1, 2, 3, 4, 5, if the repetition of digits is not allowed and numbers are divisible by either 3 or 5, is \_\_\_\_\_.

Official Ans. by NTA (32)

Sol. We need three digits numbers.

Since 
$$1 + 2 + 3 + 4 + 5 = 15$$

So, number of possible triplets for multiple of 15 is  $1 \times 2 \times 2$ 

so Ans. = 
$$4 \times |3 + 4 \times 3 - 1 \times 2 \times |2| = 32$$

8. Let  $\vec{a} = \hat{i} + 2\hat{j} - \hat{k}$ ,  $\vec{b} = \hat{i} - \hat{j}$  and  $\vec{c} = \hat{i} - \hat{j} - \hat{k}$  be three given vectors. If  $\vec{r}$  is a vector such that  $\vec{r} \times \vec{a} = \vec{c} \times \vec{a}$  and  $\vec{r} \cdot \vec{b} = 0$ , then  $\vec{r} \cdot \vec{a}$  is equal to

Official Ans. by NTA (12)

**Sol.** 
$$(\vec{r} - \vec{c}) \times \vec{a} = 0$$

$$\Rightarrow \vec{r} = \vec{c} + \lambda \vec{a}$$

Now,  $0 = \vec{b} \cdot \vec{c} + \lambda \vec{a} \cdot \vec{b}$ 

$$\Rightarrow \lambda = \frac{-\vec{b} \cdot \vec{c}}{\vec{a} \cdot \vec{b}} = -\frac{2}{-1} = 2$$

So, 
$$\vec{r} \cdot \vec{a} = \vec{a} \cdot \vec{c} + 2a^2 = 12$$

**9.** If the system of equations

$$kx + y + 2z = 1$$

$$3x - y - 2z = 2$$

$$-2x - 2y - 4z = 3$$

has infinitely many solutions, then k is equal to

Official Ans. by NTA (21)

- **Sol.** We observe  $5P_2 P_1 = 3P_3$ So, 15 - K = -6 $\Rightarrow K = 21$
- 10. The locus of the point of intersection of the lines  $(\sqrt{3})kx + ky 4\sqrt{3} = 0 \text{ and}$

$$\sqrt{3}x - y - 4(\sqrt{3})k = 0$$
 is a conic, whose

eccentricity is \_\_\_\_\_\_.

Official Ans. by NTA (2)

**Sol.** 
$$K = \frac{4\sqrt{3}}{\sqrt{3}x + y} = \frac{\sqrt{3}x - y}{4\sqrt{3}}$$

$$\Rightarrow 3x^2 - y^2 = 48$$

$$\Rightarrow \frac{x^2}{16} - \frac{y^2}{48} = 1$$

Now, 
$$48 = 16(e^2 - 1)$$

$$\Rightarrow$$
 e =  $\sqrt{4}$  = 2